# Pyrrolidinones derived from (S)-pyroglutamic acid. Part 2. ${ }^{1}$ Conformationally constrained kainoid analogues 

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#### Abstract

Novel conformationally constrained glutamate analogues are readily available from ( $S$ )-pyroglutamic acid using a bicyclic lactam as a synthetic template; diastereocontrolled modification of the pyrrolidine ring using a sequential conjugate addition-substitution strategy permits access to several kainoid analogues in a versatile strategy. The pyrrolidinone ring conformation appears to be controllable by the nature of remote substituents on the heterocyclic ring.


In the preceding paper, ${ }^{1}$ we detailed our methodology using a bicyclic template derived from ( $S$ )-pyroglutamic acid for the diastereocontrolled synthesis of C-3 substituted pyroglutaminols and pyroglutamates 1a-c. In this paper, we describe the further elaboration of these templates, leading ultimately to the 4-aryl-5-oxo and 4-arylmethyl-5-oxo analogues $\mathbf{1 d}$ of kainic



1a $\mathrm{R}^{1}=\mathrm{CH}_{2} \mathrm{OH}_{4} \mathrm{R}^{2}=\mathrm{C}\left(\mathrm{CO}_{2} \mathrm{Et}\right)_{3}, \mathrm{R}^{3}=\mathrm{CO}_{2} \mathrm{Et}$
2

1b $R^{1}=\mathrm{CH}_{2} \mathrm{OH}, \mathrm{R}^{2}=\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{BBu}, \mathrm{R}^{3}=\mathrm{CO}_{2} \mathrm{Et}$
1c $\mathrm{R}^{1}=\mathrm{CO}_{2} \mathrm{Me}, \mathrm{R}^{2}=\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{Me}, \mathrm{R}^{3}=\mathrm{H}$
1d $R^{1}=\mathrm{CO}_{2} \mathrm{Me}, \mathrm{R}^{2}=\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{Me}, \mathrm{R}^{3}=\mathrm{CH}_{2} \mathrm{Ph}$ or Ar
acid $\dagger \mathbf{2}$; ${ }^{2}$ this strategy allows for the sequential modification of C-3 and C-4 substituents of the kainoids, thereby permitting access to a diverse range of compounds. There has been considerable recent interest in the development of novel methodology to provide access to kainoids and their analogues, ${ }^{3,4}$ including organometallic ${ }^{5-7}$ and radical approaches. ${ }^{8,9}$ These analogues may be considered to be conformationally restricted forms of glutamate; such compounds are of importance as excitatory amino acid analogues ${ }^{3}$ and as conformationally controlling peptidomimetics. ${ }^{10-13}$

We expected to develop an efficient and versatile route to analogues of the kainoid group of amino acids from enone 3 (Scheme 1) using the bicyclic system to control diasteroselectivity. As described in the preceding paper, ${ }^{1}$ the C-7 ethoxycarbonyl activating group of $\mathbf{3}$ is of crucial importance for efficient conjugate addition, but it also offered considerable potential for further manipulation at C-7 after the conjugate addition step since the $\beta$-dicarbonyl system thus generated can be easily alkylated. ${ }^{14}$ Furthermore, we expected to be able to use the $\beta$-dicarbonyl moiety of $\mathbf{4}$ for $\alpha$-arylation reactions using aryllead(IV) triacetates, a strategy which has been developed in detail in recent years. ${ }^{15,16}$ The required lead(Iv) reagents are

[^0] yl)acetic acid.





4a $\mathrm{R}=\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{tBu}$
4b $\mathrm{R}=\mathrm{C}\left(\mathrm{CO}_{2} \mathrm{Et}\right)_{3}$
Scheme 1
readily available by direct plumbation of an aromatic ring or by lead/boron exchange of an arylboronic acid. Elaboration to the desired C-4 aryl functionalised pyroglutamates would then require only a short sequence of steps.

## Ring functionalisation by arylation and alkylation

Although the tricarboxylate conjugate adduct $\mathbf{4 b}$ (Scheme 1) was found not to react when subjected to phenyllead triacetate, presumably due to the substantial steric hindrance from the bulky C-6 substituent, the tert-butyloxycarbonylmethyl compound 4 a proved to be suitable for the introduction of a variety of aryl groups to give moderate to excellent combined yields of the diastereomeric products 5 and 6 (Scheme 2 and Table 1); aryllead reagents with either electron withdrawing or electron donating groups on the aromatic ring were applicable. However, the diastereomeric products 5 and 6 could be separated only with difficulty. In the case of phenyllead triacetate (generated in situ from lead(IV) tetraacetate and phenylboronic acid), the reaction proved to be sluggish, and the product was obtained in only $20 \%$ yield even after a 3 day reaction at room temperature. Optimisation of this reaction, by using an increased number of equivalents of aryllead reagent and heating the reaction at reflux for 3 days gave a greatly improved yield of $86 \%$ of the

5a $\mathrm{R}=\mathrm{Ph}$
5b $\mathrm{R}=0-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ 5c $\mathrm{R}=m-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ 5d $\mathrm{R}=p-\mathrm{MeC}_{6} \mathrm{H}_{4}$ 5e $\mathrm{R}=p-\mathrm{CF}_{3} \mathrm{C}_{6} \mathrm{H}_{4}$ $5 \mathrm{fR}=p-\mathrm{BrC}_{6} \mathrm{H}_{4}$ $\mathbf{5 g} \mathrm{R}=p-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ 5h $\mathrm{R}=\mathrm{PhCH}_{2}$

$6 \mathrm{a} R=\mathrm{Ph}$
6b R=0-MeOC $6_{6} \mathrm{H}_{4}$ 6c $\mathrm{R}=m-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ $6 d R=p-\mathrm{MeC}_{6} \mathrm{H}_{4}$ $6 \mathrm{e} \mathrm{R}=p-\mathrm{CF}_{3} \mathrm{C}_{6} \mathrm{H}_{4}$ 6f $\mathrm{R}=p-\mathrm{BrC}_{6} \mathrm{H}_{4}$ $6 \mathrm{gR}=\rho-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ 6h $\mathrm{R}=\mathrm{PhCH}_{2}$

Scheme 2
Table 1 Arylations and alkylations of lactam 4a giving products 5, 6

|  |  | Yield (\%) |  |
| :--- | :--- | :--- | :--- |
|  |  |  | $\mathbf{5}$ |
| Compound | R | $\mathbf{6}$ |  |
| $\mathbf{a}$ | Ph | 61 | 25 |
| $\mathbf{b}$ | $o-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ | 45 | 27 |
| $\mathbf{c}$ | $m-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ | 67 | 17 |
| $\mathbf{d}$ | $p-\mathrm{MeC}_{6} \mathrm{H}_{4}$ | 42 | 18 |
| $\mathbf{e}$ | $p-\mathrm{CF}_{3} \mathrm{C}_{6} \mathrm{H}_{4}$ | 50 | 22 |
| $\mathbf{f}$ | $p-\mathrm{BrC}_{6} \mathrm{H}_{4}$ | 52 | 22 |
| $\mathbf{g}$ | $p-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ | 26 | 12 |
| $\mathbf{h}$ | $\mathrm{PhCH}_{2}$ | 34 | 41 |

diastereomeric products $5 \mathbf{5}$ and $\mathbf{6 a}$ in a ratio of $2.4: 1$, and the stereochemistry of $\mathbf{5 a}$ was established by NOE studies (Fig. 1). The exo orientation of the C-7 phenyl group was determined from a set of enhancements observed between the ortho protons of this group, $\mathrm{H}-4_{\text {exo }}$ and the ortho protons of the C-2 phenyl ring. The exo orientation of the C-6 substituent was also confirmed by the observation of enhancements between the spacially proximal hydrogens $\mathrm{H}-2, \mathrm{H}-4_{\text {endo }}$, and H-6.

The reaction of $4 \mathbf{a}$ with $o$-methoxyphenyllead triacetate (generated in situ from lead(IV) tetraacetate and o-methoxyphenylboronic acid ${ }^{16}$ ) under the above conditions for 2 days gave, after careful chromatography, a mixture of unreacted starting material $\mathbf{4 a}(27 \%)$, the desired diastereomeric products $\mathbf{5 b}$ and $\mathbf{6 b}$ in $40 \%$ overall yield, and the acetate 7; $\alpha$-acetoxylation is a well documented side-reaction for lead(IV) carboxylates. ${ }^{17}$ This latter product may arise due to the greater steric


7


9a


8aR=H
$8 \mathrm{~b} R=\mathrm{Cl}$ 8c $\mathrm{R}=1$


9b


5a


5e


5h

$6 e$

5d


6d

6h

9b

Fig. 1 NOE data for selected compounds.
bulk of the $o$-methoxyphenyl group, thus allowing the competing acetoxylation process to occur. A cleaner and higher yielding arylation reaction was achieved with o-methoxyphenyllead(IV) triacetate (prepared according to the literature method ${ }^{16}$ ), giving a $72 \%$ yield of $\mathbf{5 b}$ and $\mathbf{6 b}$ in a ratio of 1.7:1. The $m$-methoxyphenyl derivatives 5 c : $\mathbf{6 c}$ were obtained in $84 \%$ yield and a ratio of $4: 1$. The $p$-methylphenyl, $p$-trifluoromethylphenyl and $p$-bromophenyl derivatives $\mathbf{5 d} \mathbf{- f}$ and $\mathbf{6 d} \mathbf{- f}$ were obtained in combined yields of 60,72 and $74 \%$ respectively as 2.3:1 ratio of diastereomers. The stereochemical assignment of these stereoisomers was possible by the presence of NOE spectroscopic enhancements from either the methylene of the C-6 substituent or $\mathrm{H}-5$ to the $o$-aromatic protons of the $\mathrm{C}-7$ sub-


Scheme 3

Table 2 Deprotections of lactams 5, 6 giving products 10, 11

| Compound | Ratio | Product | $R_{\mathrm{f}}$ | Yield <br> $(\%)$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathbf{5 a , 6} \mathbf{6 a}$ | $2.1: 1$ | $\mathbf{1 0 a}, \mathbf{1 1 a}$ | $0.24,0.32$ | 81 |
| $\mathbf{5 g ,} \mathbf{6 g}$ | $2.3: 1$ | $\mathbf{1 0 b}, \mathbf{1 1 b}$ | $0.22,0.27$ | 38 |
| $\mathbf{5 h}$ | - | $\mathbf{1 0 c}$ | 0.20 | 86 |
| $\mathbf{6 h}$ | - | $\mathbf{1 1 c}$ | 0.27 | 77 |

stituent for the diastereomers 5, or from H-6 to the same aromatic protons for diastereomer 6 (Fig. 1). However, unlike the cases outlined above, reaction with $p$-methoxyphenyllead(Iv) triacetate proved to be very problematic; the best result was obtained when lactam 4 a was converted to the corresponding enolate ( $\mathrm{NaH}-\mathrm{THF}$ ) followed by treatment with $p$-methoxyphenyllead(IV) triacetate and pyridine at rt for 5 days. Purification by column chromatography gave the products $\mathbf{5 g}$ and $\mathbf{6 g}$ in a yield of $38 \%$ as a 2.3:1 ratio.

The alkylation at C-7 of bicyclic lactam 4a was also investigated. Conversion of lactam $\mathbf{4 a}$ to the corresponding enolate (NaH-THF) followed by addition of benzyl bromide gave the products $\mathbf{5 h}$ and $\mathbf{6 h}$ in a yield of $75 \%$ as a 1:1.2 ratio. Stereochemical assignment using a series of NOE experiments as before was possible for both diastereomers of $\mathbf{5 h}$ and $\mathbf{6 h}$ (Fig. 1). Alternatively, alkylation of $\mathbf{4 a}$ with $\alpha$-iodovalerolactam $8 \mathbf{8 c}$ (readily obtained from $N$-benzylvalerolactam $\mathbf{8 a}$ in three steps via the chloride 8b) gave the diastereomeric products $\mathbf{9 a}, \mathbf{b}$ in $36 \%$ yield (ratio $1: 1$ ), and the stereochemistry of $9 \mathbf{b}$ was assigned by NOE analysis (Fig. 1).

## Conversion to pyroglutamate derivatives

Deprotection of lactams $\mathbf{5 a}, \mathbf{g}, \mathbf{h}$ and $\mathbf{6 a}, \mathbf{g}, \mathbf{h}$, by exposure to TFA in dichloromethane for 15 minutes gave the expected alcohols 10a-c and 11a-c in moderate to excellent yield (Scheme 3 and Table 2); unlike the arylated bicyclic starting materials which were separable only with difficulty by chromatography, each of the diastereomerically pure alcohols $\mathbf{1 0}$ and 11 could be easily obtained. As with the starting lactams 5 and 6, the endo aryl compounds $\mathbf{1 1}$ had greater $R_{\mathrm{f}}$ values than their C-4 epimers 10.

In order to realise our goal of developing a short synthesis of the kainoid group of amino acids using this strategy, removal of the C-7 ethoxycarbonyl substituent of intermediates $\mathbf{5 , 6}$ was required; the C-7 exo stereochemistry was desired for the most commonly occurring and most biologically active kainoids. ${ }^{4}$ Hydrolysis and decarboxylation of 10a with sodium hydroxide in ethanol at room temperature followed by heating under vacuum gave the product diastereomers 12 and 13, along with the acid 14. Further heating of this mixture effected complete decarboxylation, giving an overall yield of the products 12, 13 of $82 \%$ in a ratio of $1: 1.4$; noteworthy is the high proportion of the product $\mathbf{1 2}$ possessing the desired kainic acid configuration. The diastereomers were identified from examination of the ${ }^{1} \mathrm{H}$


Fig. 2


NMR spectrum which showed a similar distinctive splitting pattern for each diastereomer as observed for the arylated bicyclic compounds 5, $\mathbf{6}$. Thus, the signals from the methylene protons $\alpha$ to the tert-butyl ester were coincident for the trans-C$3 / 4$ compound 13, and separated by 2.7 ppm for the cis-C-3/4 compound 12. The alcohols $\mathbf{1 2}, 13$ (1:1) were treated with TFA to remove the tert-butyl group, and then oxidised using ruthenium(IV) oxide/sodium periodate and methylated with diazomethane; the expected diastereomers 15a and 16a were each obtained in $33 \%$ yield. The stereochemical assignment of 15a and 16a as shown was confirmed from NOE experiments


15a $\mathrm{R}=\mathrm{Ph}$
15b $\mathrm{R}=\mathrm{PhCH}_{2}$


17


16a $R=P h$
16b $\mathrm{R}=\mathrm{PhCH}_{2}$


18
performed on each diastereomer, the results of which are shown in Fig. 2. The diastereomer 16a showed enhancements between the H-2 and the C-6 methylene substituent, and between the $\mathrm{H}-3$ and the C-4 ortho-phenyl signals, which were indicative of the all-trans disposition of $\mathrm{H}-2,-3$ and -4 . The diastereomer 15a displayed enhancements between the $\mathrm{C}-4$ ortho-phenyl signal and that of $\mathrm{H}-2$, and the cis disposition of $\mathrm{H}-2$ and the $\mathrm{C}-3$ substituent was indicated from the enhancement of the methylene signal on irradiation of $\mathrm{H}-2$. In addition to the NOE data, the ${ }^{1} \mathrm{H}$ NMR spectra of these compounds displayed the previously discussed features for these diastereomeric arylated compounds; thus, the methylene signal was coincident for the trans-C-3/4 compound 16a and split for the cis-C-3/4 compound 15 a , which could be attributed to the shielding environment imposed upon the methylene hydrogens when positioned cis to the anisotropic phenyl ring.


19a $\mathrm{R}^{1}=\mathrm{EtO}_{2} \mathrm{C}, \mathrm{R}^{2}=\mathrm{Ph}$
19b $\mathrm{R}^{1}=\mathrm{Ph}, \mathrm{R}^{2}=\mathrm{EtO}_{2} \mathrm{C}$


21

$20 a R^{1}=H, R^{2}=P h$
$20 b R^{1}=P h, R^{2}=H$


22a $R=E t$
$22 b R=M e$

Although it was anticipated that better diastereocontrol would be obtained if the decarboxylation step was performed on the bicyclic lactams 5 and $\mathbf{6}$, in practice the initial ester hydrolysis step proved to be unreliable. Thus, hydrolysis with alcoholic sodium hydroxide of the phenyl adduct 5a, $\mathbf{6 a}(2: 1)$ gave slow ethyl ester hydrolysis and decarboxylation, along with some tert-butyl ester hydrolysis. Complete tert-butyl ester hydrolysis resulted upon repetition of this reaction at $50^{\circ} \mathrm{C}$ for $5-8 \mathrm{~h}$ to give acid $\mathbf{1 7}$ as a mixture of diastereomers at C-7. Hemiaminal ether deprotection was then easily achieved by treatment with TFA in dichloromethane for 4.5 h at room temperature to give $\mathbf{1 8}$, and alcohol oxidation was effected using ruthenium(IV) oxide-sodium periodate. The crude dicarboxylic acid product was immediately converted to the dimethyl ester ( $\mathrm{MeOH}, \mathrm{H}_{2} \mathrm{SO}_{4}$, reflux) to give the expected product as the two diastereomers 15a and 16a. However, the two minor products 19a, 19b were also isolated in low yield from this sequence, arising from incomplete hydrolysis in the initial step. Hydrolysis of the dimethyl ester 16a by treatment with sodium hydroxide in aqueous tetrahydrofuran gave the diacids 20a, b in high yield $(83 \%)$ in a ratio of $1: 3.8$, that is, favouring the trans-$\mathrm{C}-3 / 4$ arrangement. This was confirmed by re-esterification with methanol-conc. $\mathrm{H}_{2} \mathrm{SO}_{4}$ which gave a $1: 5$ diastereomeric mixture of the dimethyl ester product 15a and 16a.

The C-4 disubstituted derivative 21 was readily obtained using the above oxidation-esterification sequence on pyrrolidinone 10a in good yield ( $41 \%$ ), but under the same conditions the alcohol 11a gave no product.

Attempted hydrolysis and decarboxylation of the benzyl compound 6 h proved to be problematic. Because of the easier hydrolysis which had been observed for lactams 10a and 11a, attention was turned towards the hydrolysis of the readily separable benzyl compounds 10c and 11c. Selective ethyl ester hydrolysis of $\mathbf{1 1 c}$ with $\mathrm{NaOH}(1 \mathrm{M})$, treatment with TFA then with $\mathrm{RuO}_{2}-\mathrm{NaIO}_{4}$ and finally with diazomethane, gave the dimethyl monoethyl ester and trimethyl ester products 22a and 22b in yields of $23 \%$ and $25 \%$ respectively. This result shows that some ethyl ester hydrolysis had occurred, but highlights the sterically hindered nature of the C-4 esters in substrates of this type.

A similar sequence was used for compound 10c; hydrolysis with NaOH gave the carboxylic acid 23 in a yield of $99 \%$.


23
Decarboxylation (heating at $135^{\circ} \mathrm{C}$ at 0.5 mbar ), deprotection (TFA), oxidation and esterification (diazomethane) gave the completely separated product diastereomers $\mathbf{1 5 b}$ and 16b in

Table 3 Coupling constants and dihedral angles

| Compound | Coupling constant/ <br> $\mathrm{Hzz}^{\text {a }}$ | $\mathrm{C}(2) \mathrm{H}-\mathrm{C}(3) \mathrm{H}$ <br> Dihedral angle/ $/$ |
| :--- | :--- | :--- |
| $\mathbf{1 5 a}(\mathrm{R}=\mathrm{H})$ | $6.0(6.5)$ | 118 |
| $\mathbf{1 5 a}$ | $5.5(5.5)$ | 117 |
| $\mathbf{1 6 a}$ | $6.5(6.5)$ | 121 |
| $\mathbf{1 5 b}$ | $8.0(8.0)$ | 153 |
| $\mathbf{1 6 b}$ | $2.5(2.5)$ | 125 |
| ${ }^{\text {a }}{\mathrm{In} \mathrm{CDCl}_{3}\left(\mathrm{C}_{6} \mathrm{D}_{6}\right) .}$ |  |  |

Table 4 Selected ${ }^{1} \mathrm{H}$ NMR spectroscopic data for compounds 5, 6
$\begin{array}{llllll}\hline & \delta & & & & \\$\cline { 2 - 5 } Compound \& \& $\left.\mathrm{H}-2 & \mathrm{H}-4_{\text {endo }} & \mathrm{H}-4_{\text {exo }} & \mathrm{CH}_{2} \mathrm{CO}^{\mathrm{t}} \mathrm{Bu}\end{array}\right) R_{\mathrm{f}}$.
overall yields of 7 and $31 \%$ respectively from the alcohol $\mathbf{1 0 c}$. The ${ }^{1} \mathrm{H}$ NMR spectrum of the trans-C- $3 / 4$ compound $\mathbf{1 6 b}$ was amenable to NOE analysis for stereochemical determination, and the significant enhancements are shown in Fig. 2. Enhancements between H-4, the methylene protons of the C-3 substituent, and H-2 indicated their cis relationship; this stereochemistry would be expected for the major diastereomer on consideration of steric interactions.

## Conformational studies

This synthetic route represents a novel and simple, but potentially generalisable approach to highly functionalised pyrrolidinones, and is complementary to existing literature protocols. In particular, it provides access to novel pyroglutamate analogues of the kainoid group of amino acids possessing substituents with $\pi$-electron density at C-4; the synthesis of conformationally constrained pyroglutamates has attracted recent attention. ${ }^{18}$ It was apparent from examination of the ${ }^{1} \mathrm{H}$ NMR spectra of the final dimethyl ester pyroglutamates $\mathbf{1 5 a}$, $\mathbf{b}$ and 16a, $\mathbf{b}$, as well as the unsubstituted derivative $\mathbf{1 5}(\mathrm{R}=\mathrm{H}),{ }^{1}$ that the coupling constants between the $\mathrm{H}-2$ and $\mathrm{H}-3$ varied between diastereomers and with the type of C-4 substituent (Table 3); molecular modelling studies of each of these compounds ${ }^{19}$ confirmed these experimental findings, since the $\mathrm{H}-2-$ $\mathrm{H}-3$ dihedral angle was found to vary depending on the nature of the C-4 substituent, and appears to be greatest for the more sterically congested C-4 (aryl) series of compounds $\mathbf{6 a}, \mathbf{b}$. Thus, it would appear that analogues of well defined glutamate conformers could be available by variation in the C-4 substituent of compounds of type 15 and 16.

## NMR spectroscopy

In common with the simpler systems discussed in the preceding paper, a consistent pattern was observed with regard to chemical shifts and coupling constant values in the NMR spectra (Table 4). Thus, for the aryl adducts $5 \mathbf{5 a - e}$ and $\mathbf{6 a - e}, \mathrm{H}-4_{\text {endo }}$ has a lower chemical shift than $\mathrm{H}-4_{\text {exo }}$ with $\Delta \delta$ of about 0.4 ; how-
ever, $\mathbf{6} \mathbf{h}$ is exceptional, and this difference is more than double, a result which can be attributed to the anisotropy of the adjacent benzyl substituent. Similar anisotropy is responsible for the well separated resonances of the C-6 methylene substituent protons of the exo aryl diastereomers $\mathbf{5}$, but coincident resonance in endo aryl 6.

## Experimental

For general experimental procedures, see our earlier reports. ${ }^{14,20}$

## General arylation methods

Method 1. To a mixture of lead(Iv) tetraacetate, arylboronic acid and mercury(II) acetate in an inert atmosphere was added ethanol-free chloroform and the mixture was then stirred at $40^{\circ} \mathrm{C}$ for 2 h . A solution of the lactam 4 a in chloroform was then added and the mixture heated at reflux for 72 h . After cooling to rt the mixture was filtered through Celite ${ }^{\circledR}$, washed with chloroform, and the organic layer washed with sulfuric acid ( $10 \%$ aq.). The aqueous phase was washed twice with chloroform and the combined organic phases were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered and evaporated in vacuo prior to purification.

Method 2. To a solution of the lactam $\mathbf{4 a}$ in ethanol-free chloroform was added pyridine and the aryllead(Iv) complex. The mixture was then heated at reflux under an argon atmosphere for the specified period. After cooling, the mixture was worked-up as above.

Method 3. To a stirred solution of the lactam 4a in THF at rt was added sodium hydride ( 1.1 eq.). After 1 h this solution was transferred to a solution of the aryllead(Iv) complex (1.5 eq.) in pyridine ( 2 ml ), using THF ( 1 ml ) to rinse the flask out. The mixture was then stirred at room temperature. After 3 days another aliquot of pyridine ( 2 ml ) and aryllead(iv) complex ( 1.2 eq.) was added. The mixture was then worked-up after a further 2 days as above.

Method 4. A solution of lactam $\mathbf{4 a}$ and the required aryllead compound ${ }^{15,21}$ ( 1.05 eq.) was heated under reflux in chloroform $(10 \mathrm{ml})$ with pyridine ( 3 eq. ) for 72 h . The reaction mixture was cooled to rt , diluted with chloroform ( 10 ml ), washed with 2 M $\mathrm{HCl}(10 \mathrm{ml})$ and water $(15 \mathrm{ml})$, dried $\left(\mathrm{MgSO}_{4}\right)$ and the solvent removed in vacuo.

## ( $2 R, 5 S, 6 S, 7 R$ )- and ( $2 R, 5 S, 6 S, 7 S$ )-6-tert-Butoxycarbonyl-methyl-7-ethoxycarbonyl-3-oxa-8-oxo-2,7-diphenyl-1-azabicyclo[3.3.0]octanes 5a and 6a

According to Method 1 , lactam $4 \mathrm{a}(0.30 \mathrm{~g}, 0.77 \mathrm{mmol})$ was reacted with phenylboronic acid ( $0.19 \mathrm{~g}, 1.5 \mathrm{mmol}$ ), lead(Iv) tetraacetate ( $0.72 \mathrm{~g}, 1.5 \mathrm{mmol}$ ), mercury(II) acetate ( $49 \mathrm{mg}, 0.15$ mmol ), and pyridine ( $0.37 \mathrm{~g}, 4.62 \mathrm{mmol}$ ) in $\mathrm{CHCl}_{3}(30 \mathrm{ml})$ at reflux for 3 days to give after work-up and purification using flash column chromatography $\left(\mathrm{Et}_{2} \mathrm{O}\right.$-cyclohexane, 1:3) the product as a colourless glass of diastereomeric ratio 5a:6a of 2.4:1 ( $0.31 \mathrm{~g}, 86 \%$ ).

Data for 5a. $R_{\mathrm{f}} 0.19 ; v_{\max }($ film $) / \mathrm{cm}^{-1} 2980(\mathrm{~s}), 1368(\mathrm{~m}), 1245$ (s), $1154(\mathrm{~s}), 1050(\mathrm{~m}), 1028(\mathrm{~m}), 701(\mathrm{~m}), 665(\mathrm{~m}) ; \delta_{\mathrm{H}}(500$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.31\left(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 1.41(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.64\left(1 \mathrm{H}, \mathrm{dd}, J 16.5\right.$ and $\left.11.5, \mathrm{C} H \mathrm{HCO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 2.42$ $\left(1 \mathrm{H}, \mathrm{dd}, J 16.5\right.$ and $\left.4.0, \mathrm{CH} H \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 3.58-3.62(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6)$, $3.69-3.74(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5), 3.97\left(1 \mathrm{H}, \mathrm{dd}, J 9.0\right.$ and $\left.7.5, \mathrm{H}-4_{\text {endo }}\right)$, $4.35\left(2 \mathrm{H}, \mathrm{q}, J 7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 4.43(1 \mathrm{H}, \mathrm{dd}, J 9.0$ and 6.0 , $\mathrm{H}-\mathrm{4}_{\text {exo }}$ ), $6.43(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-2), 7.11-7.13$ ( $2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H$ ), $7.30-7.45$ $(6 \mathrm{H}, \mathrm{m}, \operatorname{Ar} H), 7.57-7.59(2 \mathrm{H}, \mathrm{m}, \operatorname{Ar} H) ; \delta_{\mathrm{C}}(125.8 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 14.01\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 28.00\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 35.99\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right)$, $44.49(\mathrm{C}-6), 61.32(\mathrm{C}-5), 62.40\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 69.72(\mathrm{C}-7)$ and $72.68(\mathrm{C}-4), 81.28\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right)}\right) 86.99(\mathrm{C}-2), 126.11,127.95$,
128.03, 128.46, 128.60 and $128.81(\mathrm{ArCH}), 134.10(\mathrm{ArC})$, $138.37(\mathrm{ArC}), 169.90,170.68$ and $170.88(3 \times \mathrm{CO})$.

Data for 6a. $R_{\mathrm{f}} 0.21 ; v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 2980(\mathrm{~s}), 1368(\mathrm{~m}), 1245$ (s), $1154(\mathrm{~s}), 1050(\mathrm{~m}), 1028(\mathrm{~m}), 701(\mathrm{~m}), 665(\mathrm{~m}) ; \delta_{\mathrm{H}}(500$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.25\left(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 1.44(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 2.70-2.78\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 3.08-3.12(1 \mathrm{H}, \mathrm{m}$, $\mathrm{H}-6), 3.90\left(1 \mathrm{H}, \mathrm{dd}, J 9.0\right.$ and $\left.7.5, \mathrm{H}-4_{\text {endo }}\right), 4.12(1 \mathrm{H}, \mathrm{dd}, J 14.0$ and 7.5, H-5), 4.21-4.33 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2}$ ), $4.35(1 \mathrm{H}, \mathrm{dd}, J 9.0$ and $\left.6.0, \mathrm{H}-4_{\text {exo }}\right), 6.37(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-2), 7.33-7.44(8 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H)$, $7.48-7.50(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 14.00$ $\left.\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 28.03\left(\mathrm{C}_{\left(\mathrm{CH}_{3}\right)}\right)_{3}\right), 34.48\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 47.38(\mathrm{C}-6)$, 62.00 and $62.16\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right.$ and $\left.\mathrm{C}-5\right)$, $68.41(\mathrm{C}-7)$ and 72.48 (C-4), $81.54\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 86.53(\mathrm{C}-2), 126.12,128.00,128.46 \text { and }}\right.$ $128.74(\mathrm{ArCH}), 135.41(\mathrm{ArC}), 137.98(\mathrm{ArC}), 169.42,170.28$ and $171.09(3 \times \mathrm{CO}) ; m / z\left(\mathrm{CI}\left(\mathrm{NH}_{3}\right)\right) 466\left(\mathrm{M}+\mathrm{H}^{+}, 100 \%\right)$; HRMS 466.2230, $\mathrm{C}_{27} \mathrm{H}_{32} \mathrm{NO}_{6}$ requires 466.2231.

## ( $2 R, 5 S, 6 S, 7 R$ )- and ( $2 R, 5 S, 6 S, 7 S$ )-6-tert-Butoxycarbonyl-methyl-7-ethoxycarbonyl-7-o-methoxyphenyl-3-oxa-8-oxo-2-phenyl-1-azabicyclo[3.3.0]octanes 5b and 6b

According to Method 1, lactam $4 \mathrm{a}(73 \mathrm{mg}, 0.19 \mathrm{mmol}$ ) was reacted with 2-methoxyphenylboronic acid ( $59 \mathrm{mg}, 0.38 \mathrm{mmol}$ ), lead(IV) tetraacetate ( $0.18 \mathrm{~g}, 0.38 \mathrm{mmol}$ ), mercury(II) acetate ( $12 \mathrm{mg}, 0.04 \mathrm{mmol}$ ), and pyridine ( $89 \mathrm{mg}, 1.1 \mathrm{mmol}$ ) in $\mathrm{CHCl}_{3}$ $(10 \mathrm{ml})$ at reflux for 2 days. Work-up and purification by flash column chromatography (EtOAc-DCM, 50:1) gave several products, including lactam $\mathbf{4 a}(20 \mathrm{mg}, 27 \%)$, the aryl product $\mathbf{5 b}$ and the acetate product $7(4 \mathrm{mg}, 5 \%)$.

According to Method 2, o-methoxyphenyllead(iv) triacetate ( $0.22 \mathrm{~g}, 0.44 \mathrm{mmol}$ ) and the lactam $\mathbf{4 a}(86 \mathrm{mg}, 0.22 \mathrm{mmol})$ were reacted in pyridine ( $87 \mathrm{mg}, 1.1 \mathrm{mmol}$ ) and $\mathrm{CHCl}_{3}(6 \mathrm{ml})$ at reflux for 3 days. Work-up gave the crude mixture as a yellow glass which was shown to be a mixture of product diastereomers $\mathbf{5 b}, \mathbf{6 b}(1.7: 1)$, along with some starting lactam $\mathbf{4 a}$, which was purified by flash column chromatography ( $\mathrm{DCM}-\mathrm{MeOH}-$ $\mathrm{Et}_{3} \mathrm{~N}, 200: 1: 1$ ) to give the product as incompletely separable diastereoisomers ( $79 \mathrm{mg}, 72 \%$ ).

Data for 5b. $R_{\mathrm{f}} 0.11$ ( $\left.\mathrm{Et}_{2} \mathrm{O}-\operatorname{Petrol}(30-40), 1: 2\right) ; v_{\text {max }}(\mathrm{film}) /$ $\mathrm{cm}^{-1} 1718(\mathrm{~s}), 1493(\mathrm{~m}), 1459(\mathrm{~m}), 1367(\mathrm{~m}), 1308(\mathrm{~m}), 1248(\mathrm{~s})$, $1154(\mathrm{~s}), 1026(\mathrm{~m}), 755(\mathrm{~m}), 700(\mathrm{~m}) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.24$ $\left(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 1.40\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.71(1 \mathrm{H}, \mathrm{dd}$, $J 16.5$ and $\left.12.0, \mathrm{CHHCO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 2.70(1 \mathrm{H}, \mathrm{dd}, J 16.5$ and 3.5 , $\left.\mathrm{CH} H \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 3.69-3.73(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5), 3.76-3.82(4 \mathrm{H}, \mathrm{m}, \mathrm{H}-6$ and $\left.\mathrm{OCH}_{3}\right), 3.98\left(1 \mathrm{H}, \mathrm{t}, J 8.5, \mathrm{H}-4_{\text {endo }}\right), 4.23-4.30(2 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 4.42\left(1 \mathrm{H}, \mathrm{dd}, J 8.5\right.$ and $\left.6.0, \mathrm{H}-4_{\text {exo }}\right), 6.38(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-2)$, 6.89-6.94 (2H, m, $\operatorname{Ar} H), 7.12-7.14(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.29-7.32$ $(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.35-7.43(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.56-7.58(2 \mathrm{H}, \mathrm{m}$, $\mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 14.05\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 28.00$ $\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right)$, $34.76\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 43.12(\mathrm{C}-6), 55.07\left(\mathrm{OCH}_{3}\right)$, $61.76(\mathrm{C}-5), 61.92\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 67.31(\mathrm{C}-7), 72.69(\mathrm{C}-4), 80.98$ $\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 87.02(\mathrm{C}-2), 111.24(\mathrm{ArCH}), 121.53(\mathrm{ArCH}), 123.81}\right.$ $(\mathrm{ArCH}), 126.07(\mathrm{ArCH}), 128.34(\mathrm{ArCH}), 128.50(\mathrm{ArCH})$, $128.72(\mathrm{ArCH}), 129.50(\mathrm{C}-16), 138.54(\mathrm{ArC}), 156.90(\mathrm{ArC})$, $\left.170.02\left(\mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 171.47(\mathrm{C}(\mathrm{O}) \mathrm{N})+\mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right) ; \mathrm{m} / \mathrm{z}$ $\left(\mathrm{APCI}^{+}\right) 496\left(\mathrm{M}+\mathrm{H}^{+}, 21 \%\right), 440(100)$.

Data for 6b. $R_{\mathrm{f}} 0.30$ (DCM-EtOAc, $50: 1$ ); $[a]_{\mathrm{D}}^{26}+148.4$ (c 0.31 in $\mathrm{CHCl}_{3}$ ); $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 1720(\mathrm{~s}), 1494$ (m), 1461 (m), 1368 (m), 1309 (m), 1252 (s), 1224 (s), 1154 (s), 1026 (m), 755 $(\mathrm{m}), 699(\mathrm{~m}) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.25(3 \mathrm{H}, \mathrm{t}, J 7.0$, $\left.\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 1.41\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 2.63(1 \mathrm{H}, \mathrm{dd}, J 17.0$ and 3.0 , $\left.\mathrm{CH} \mathrm{HCO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 2.93\left(1 \mathrm{H}, \mathrm{dd}, J 17.0\right.$ and $\left.11.5, \mathrm{CHHCO}{ }^{\mathrm{t}} \mathrm{Bu}\right)$, $\left.3.02-3.06(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6), 3.84(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH})_{3}\right), 3.90(1 \mathrm{H}, \mathrm{dd}, J 8.5$ and $\left.7.0, \mathrm{H}-4_{\text {endo }}\right), 4.11-4.15(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5), 4.20-4.33(3 \mathrm{H}, \mathrm{m}$, $\mathrm{CH}_{3} \mathrm{CH}_{2}$ and $\left.\mathrm{H}-4_{\text {exo }}\right), 6.41(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-2), 6.97(1 \mathrm{H}, \mathrm{d}, J 8.0$, $\mathrm{Ar} H), 7.00-7.03(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.32-7.39(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.46-$ $7.48(2 \mathrm{H}, \mathrm{m}, \mathrm{ArH}) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 14.33\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right)$, $28.02\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 34.32\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 47.54(\mathrm{C}-6), 55.71$
$\left(\mathrm{OCH}_{3}\right), 61.62\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 61.85(\mathrm{C}-5), 66.67(\mathrm{C}-7), 72.52(\mathrm{C}-4)$, $81.06\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 86.76(\mathrm{C}-2), 112.26(\mathrm{ArCH}), 121.09(\mathrm{ArCH}) \text {, }}\right.$ $125.57(\mathrm{ArCH}), 126.16(\mathrm{ArCH}), 128.39(\mathrm{ArCH}), 128.58$ $(\mathrm{ArCH}), 128.77(\mathrm{ArCH}), 129.35(\mathrm{ArCH}), 138.25(\mathrm{ArC}), 157.28$ $(\mathrm{ArC}), 169.09 \quad(\mathrm{C}(\mathrm{O}) \mathrm{N}), 170.87 \quad\left(\mathrm{CO}_{2} \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 171.69$ $\left(\mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right) ; m / z\left(\mathrm{APCI}^{+}\right) 496\left(\mathrm{M}+\mathrm{H}^{+}, 33 \%\right), 440(100), 262$ (57); HRMS ( $\mathrm{CI}^{+}$) 496.2335, $\mathrm{C}_{28} \mathrm{H}_{33} \mathrm{NO}_{7}\left(\mathrm{M}+\mathrm{H}^{+}\right)$requires 496.2335.

Data for acetate 7. $R_{\mathrm{f}} 0.30$ (EtOAc-petrol (40-60), 1:2); $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 1728(\mathrm{~s}), 1370(\mathrm{~m}), 1231(\mathrm{~m}), 1155(\mathrm{~m}) ; \delta_{\mathrm{H}}(500$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.31\left(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 1.45(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 2.23\left(3 \mathrm{H}, \mathrm{s}, \mathrm{COCH}_{3}\right), 2.51(1 \mathrm{H}, \mathrm{dd}, J 17.0$ and 11.0 , $\left.\mathrm{CHHCO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 2.86\left(1 \mathrm{H}, \mathrm{dd}, J 17.0\right.$ and $\left.3.5, \mathrm{CHHCO}{ }_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right)$, 3.12-3.16 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6$ ), 3.88-3.93 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-4 \mathrm{H}$ and $\mathrm{H}-5$ ), 4.27-4.37 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2}$ ), 4.37-4.42 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-4 \mathrm{H}$ ), 6.33 ( $1 \mathrm{H}, \mathrm{s}, \mathrm{H}-2$ ), 7.33-7.39 (3H, m, $\mathrm{Ar} H), 7.44-7.46(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H)$; $\delta_{\mathrm{C}}\left(125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 13.98\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 21.07\left(\mathrm{COCH}_{3}\right)$, $28.02\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 34.71\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 44.68(\mathrm{C}-6), 61.27(\mathrm{C}-5) \text {, }}\right.$ $62.63\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 72.58(\mathrm{C}-4), 81.69\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 87.16(\mathrm{C}-2) \text {, }}\right.$ 88.07 (C-7), 126.14, 128.49 and $128.86(\mathrm{ArCH}), 137.69(\mathrm{ArC})$, 165.19, 166.63, 169.37 and $170.57(4 \times \mathrm{CO}) ; \mathrm{m} / \mathrm{z}\left(\mathrm{CI}\left(\mathrm{NH}_{3}\right)\right) 448$ $\left(\mathrm{M}+\mathrm{H}^{+}, 92 \%\right), 392$ (69); HRMS ( $\mathrm{CI}^{+}$) 448.1971, $\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{NO}_{8}$ $\left(\mathrm{M}+\mathrm{H}^{+}\right)$requires 448.1971 .

## ( $2 R, 5 S, 6 S, 7 R$ )- and ( $2 R, 5 S, 6 S, 7 S$ )-6-tert-Butoxycarbonyl-methyl-7-ethoxycarbonyl-7-m-methoxyphenyl-3-oxa-8-oxo-2-phenyl-1-azabicyclo[3.3.0]octanes 5 c and $\mathbf{6 c}$

According to Method 4, lactam $\mathbf{4 a}(118 \mathrm{mg}, 0.30 \mathrm{mmol})$ and $m$-methoxyphenyllead triacetate ${ }^{21}(161 \mathrm{mg}, 0.42 \mathrm{mmol})$ were reacted together to give a pale yellow oil ( $126 \mathrm{mg}, 84 \%$ ). The two diastereomers $5 \mathbf{c}$ and $\mathbf{6 c}$ were present in the ratio $4: 1$ but were not separable by flash column chromatography.

Data for 5c. $101 \mathrm{mg}, 67 \% ; R_{\mathrm{f}} 0.36$ (petrol-EtOAc, 3:1); $v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 2982(\mathrm{~m}), 1720(\mathrm{br}, \mathrm{s}), 1515(\mathrm{~m}) ; \delta_{\mathrm{H}}(400$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.32\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.41(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.66\left(1 \mathrm{H}, \mathrm{dd}, J 16.5\right.$ and $\left.11.5, \mathrm{CHCO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 2.45(1 \mathrm{H}$, dd, $J 16.5$ and $\left.3.7, \mathrm{CHCO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 3.54-3.58(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6), 3.60$ $\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 3.73\left(1 \mathrm{H}, \mathrm{dd}, J 13.8\right.$ and $\left.7.7, \mathrm{H}-4_{\text {endo }}\right), 3.85-3.95$ $(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5), 4.35\left(2 \mathrm{H}, \mathrm{q}, J 7.1, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.44-4.48(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{H}-4_{\text {exo }}\right), 6.43(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-2), 6.64-7.61(9 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H) ; \delta_{\mathrm{C}}(50.3$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 13.91\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 27.91\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 35.74$ $\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{H}} \mathrm{Bu}\right), 44.29(\mathrm{C}-6), 54.96\left(\mathrm{ArOCH}_{3}\right), 61.11(\mathrm{C}-5), 62.40$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 69.70(\mathrm{C}-7), 72.89(\mathrm{C}-4), 81.32\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 86.67}\right.$ (C-2), 113.02, 114.15, 120.60, 126.23, 128.78 and 129.03 $(\mathrm{ArCH}), 135.6,138.75$ and $160.12(\mathrm{ArC}), 170.0,170.2$ and $171.29(\mathrm{CO}) ; \mathrm{m} / \mathrm{z}\left(\mathrm{CI}, \mathrm{NH}_{3}\right) 440(52 \%)$; HRMS 496.2335. $\mathrm{C}_{28} \mathrm{H}_{33} \mathrm{NO}_{6}\left(\mathrm{M}+\mathrm{H}^{+}\right)$requires 496.5485.

Data for $6 \mathrm{c} .25 \mathrm{mg}, 17 \% ; R_{\mathrm{f}} 0.36$ (petrol-EtOAc, 3:1); $v_{\text {max }} /$ $\mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 2982(\mathrm{~m}), 1720(\mathrm{~s}), 1515(\mathrm{~m}) ; \delta_{\mathrm{H}}(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 1.25\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.44\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right)$, 2.74-2.76 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}$ ), $3.08-3.11(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6), 3.82$ $\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 3.90\left(1 \mathrm{H}, \mathrm{dd}, J 8.9\right.$ and $\left.7.3, \mathrm{H}-4_{\text {endo }}\right), 4.12(1 \mathrm{H}$, $\mathrm{dt}, J 7.2$ and $7.2, \mathrm{H}-5), 4.28\left(2 \mathrm{H}, \mathrm{q}, J 7.1, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.32-4.38$ $\left(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-4_{e x}\right), 6.37(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-2), 6.88-7.60(9 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H)$; $\delta_{\mathrm{C}}\left(50.3 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 13.91\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 27.91\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 34.5}\right.$ $\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 47.9(\mathrm{C}-6), 55.2\left(\mathrm{ArOCH}_{3}\right), 61.9(\mathrm{C}-5), 62.05$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 68.2(\mathrm{C}-7), 72.2(\mathrm{C}-4), 81.5\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 86.6(\mathrm{C}-2), ~}^{\text {, }}\right.$ 113.1, 114.2, 120.2, 128.7, 129.0 and $129.8(\mathrm{ArCH}), 137.0$, 138.1 and $159.9(\mathrm{ArC}), 169.7,170.2$ and $171.3(\mathrm{CO}) ; m / z(\mathrm{CI}$, $\left.\mathrm{NH}_{3}\right) 440$ (52\%); HRMS 496.2335. $\mathrm{C}_{28} \mathrm{H}_{33} \mathrm{NO}_{6}\left(\mathrm{M}+\mathrm{H}^{+}\right)$ requires 496.5485 .

## ( $2 R, 5 S, 6 S, 7 R$ )- and ( $2 R, 5 S, 6 S, 7 S$ )-7-Ethoxycarbonyl-6-tert-butoxycarbonylmethyl-7-p-methylphenyl-3-oxa-8-oxo-2-phenyl-1-azabicyclo[3.3.0]octanes 5d and 6d

According to Method 4, lactam $\mathbf{4 a}(156 \mathrm{mg}, 0.40 \mathrm{mmol})$ was reacted with $p$-methylphenyllead triacetate ${ }^{21}$ ( $200 \mathrm{mg}, 0.55$
mmol ) to yield the title compound as two separate diastereomers $\mathbf{5 d}$ and $\mathbf{6 d}$ in a ratio of 2.3:1 as yellow oils ( $115 \mathrm{mg}, 60 \%$ ).

Data for 5d. $81 \mathrm{mg}, 42 \% ; R_{\mathrm{f}} 0.38$ (petrol-EtOAc, 3:1); $v_{\max }($ film $) / \mathrm{cm}^{-1} 2980(\mathrm{~m}), 2933(\mathrm{~m}), 1724$ (br, s), 1515 (w); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.30\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.44(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.66\left(1 \mathrm{H}, \mathrm{dd}, J 16.4\right.$ and $\left.11.2, \mathrm{CHCO}_{2}{ }^{\mathrm{B}} \mathrm{Bu}\right), 2.33(3 \mathrm{H}$, $\left.\mathrm{s}, \mathrm{ArCH})_{3}\right), 2.43,\left(1 \mathrm{H}, \mathrm{dd}, J 16.4\right.$ and $\left.3.8, \mathrm{CHCO}_{2}{ }^{\text {t }} \mathrm{Bu}\right), 3.53-3.62$ $(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6), 3.70\left(1 \mathrm{H}, \mathrm{dd}, J 13.7\right.$ and $\left.7.5, \mathrm{H}-4_{\text {endo }}\right), 3.86-4.00$ $(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5), 4.33\left(2 \mathrm{H}, \mathrm{q}, J 7.0, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.42-4.45(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{H}-4_{\text {exo }}\right), 6.42(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-2), 7.00(2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar} H), 7.14(2 \mathrm{H}, \mathrm{d}$, $J 8.2, \mathrm{ArH}), 7.34-7.60(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), \delta_{\mathrm{C}}\left(100.7 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $14.05\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 20.85\left(\mathrm{ArCH}_{3}\right), 26.78\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 34.36$ $\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\text {t }} \mathrm{Bu}\right), 47.23(\mathrm{C}-6), 61.95\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 62.16(\mathrm{C}-5), 68.04$ (C-7), $72.57(\mathrm{C}-4), 81.30\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 86.38(\mathrm{C}-2), 126.01,127.74 \text {, }}^{\text {, }}\right.$ 128.67, 129.03 and $132.34(\mathrm{ArCH}), 137.67$ and $137.96(\mathrm{ArC})$, 169.45, 170.34 and $170.71(\mathrm{CO}) ; m / z\left(\mathrm{CI}, \mathrm{NH}_{3}\right) 480\left(\mathrm{MH}^{+}\right.$, $30 \%$ ), 424 (100); HRMS 480.2385. $\mathrm{C}_{28} \mathrm{H}_{33} \mathrm{NO}_{6}\left(\mathrm{M}+\mathrm{H}^{+}\right)$ requires 480.2386 .

Data for 6d. $34 \mathrm{mg}, 18 \% ; R_{\mathrm{f}} 0.38$ (petrol-EtOAc, 3:1); $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 2980(\mathrm{~m}), 2933(\mathrm{~m}), 1724$ (br, s), 1515 (w); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.30\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.41(9 \mathrm{H}$, $\left.\mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 2.36\left(3 \mathrm{H}, \mathrm{s}, \mathrm{ArCH}_{3}\right), 2.72,\left(2 \mathrm{H}, \mathrm{d}, J 7.6, \mathrm{CH}_{2}-\right.$ $\left.\mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 3.08-3.12(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6), 3.86-3.90\left(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-4_{\text {endo }}\right)$, 4.01-4.15 (1H, m, H-5), 4.24-4.29 (1H, m, H-4 exo ), 4.34 ( $2 \mathrm{H}, \mathrm{q}$, $J 7.1, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 6.36 ( $1 \mathrm{H}, \mathrm{s}, \mathrm{H}-2$ ), $7.20-7.60(9 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ); $\delta_{\mathrm{C}}\left(100.7 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 13.66\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 20.27\left(\mathrm{ArCH}_{3}\right), 27.43$
 $\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 69.02(\mathrm{C}-7), 72.10(\mathrm{C}-4), 80.45\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right)}\right), 86.59$ (C-2), 125.79, 127.61, 127.96, 128.11, 128.32 and 128.97 $(\mathrm{ArCH}), 137.23$ and $138.36(\mathrm{ArC}), 169.52,170.53$ and 170.23 (CO); $m / z\left(\mathrm{CI}, \mathrm{NH}_{3}\right) 480\left(\mathrm{MH}^{+}, 30 \%\right), 424$ (100); HRMS 480.2385. $\mathrm{C}_{28} \mathrm{H}_{33} \mathrm{NO}_{6}\left(\mathrm{MH}^{+}\right)$requires 480.2386.

## $(2 R, 5 S, 6 S, 7 R)$ - and ( $2 R, 5 S, 6 S, 7 S)$-6-tert-Butoxycarbonyl-methyl-7-ethoxycarbonyl-3-oxa-8-oxo-2-phenyl-7-p-trifluoro-methylphenyl-1-azabicyclo[3.3.0]octanes 5 e and 6 e

According to Method 4, lactam $\mathbf{4 a}(106 \mathrm{mg}, 0.27 \mathrm{mmol})$ was reacted with $p$-trifluoromethylphenyllead triacetate ${ }^{21}(151 \mathrm{mg}$, 0.36 mmol ) to yield two separate diastereomers $\mathbf{5 e}$ e, $\mathbf{6 e}$ in a 2.3:1 ratio ( $104 \mathrm{mg}, 72 \%$ ).

Data for 5e. Yellow oil; $72 \mathrm{mg}, 50 \% ; R_{\mathrm{f}} 0.39$ (petrol-EtOAc, 3:1); $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 3019(\mathrm{br}, \mathrm{m}), 2930(\mathrm{~m}), 1721(\mathrm{br}, \mathrm{s}), 1515$ (w); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.31\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.41$ $\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.64\left(1 \mathrm{H}, \mathrm{dd}, J_{1} 16.4, J_{2} 10.7, \mathrm{CHCO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right)$, $2.39\left(1 \mathrm{H}, \mathrm{dd}, J_{1} 16.4, J_{2} 4.0, \mathrm{CHCO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 3.63-3.74(2 \mathrm{H}, \mathrm{m}$, $\mathrm{H}-4_{\text {endo }}$ and $\left.\mathrm{H}-6\right), 3.97-4.01(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5), 4.35(2 \mathrm{H}, \mathrm{q}, J 7.1$, $\left.\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.42-4.46\left(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-4_{\text {exo }}\right), 6.42(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-2), 7.26$ ( $2 \mathrm{H}, \mathrm{d}, J 7.2, \mathrm{Ar} H$ ), $7.38-7.47$ (3H, m, $\operatorname{ArH}$ ), 7.56 ( $2 \mathrm{H}, \mathrm{m}$, $\mathrm{Ar} H), 7.61(2 \mathrm{H}, \mathrm{d}, J 7.2, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(100.7 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 13.98$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 27.96\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 35.80\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 44.32(\mathrm{C}-6) \text {, }}\right.$ $61.35(\mathrm{C}-5), 62.72\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 69.66(\mathrm{C}-7), 72.56(\mathrm{C}-4), 81.58$ $\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 87.03(\mathrm{C}-2), 125.72,125.75,126.02,128.51,128.64 \text {, }}^{\text {, }}\right.$ 128.67 and $128.99(\mathrm{ArCH}), 138.09$ and $138.10(\mathrm{ArC}), 169.21$, 170.12 and $170.43(\mathrm{CO}) ; m / z\left(\mathrm{CI}, \mathrm{NH}_{3}\right) 534\left(\mathrm{MH}^{+}, 12 \%\right), 478$ (100); HRMS 534.2097. $\mathrm{C}_{28} \mathrm{H}_{33} \mathrm{NO}_{6}\left(\mathrm{M}+\mathrm{H}^{+}\right)$requires 534.2103.

Data for 6e. Colourless oil; 22\%; $R_{\mathrm{f}} 0.49$ (petrol-EtOAc, 3:1); $v_{\text {max }}$ (film)/ $\mathrm{cm}^{-1} 3019$ (br), $2930(\mathrm{~m}), 1721$ (br, s), 1515 (w); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}\right) 0.69\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{2} \mathrm{CH}_{2}\right), 1.24(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 2.69-2.81,\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{C}} \mathrm{Bu}\right), 2.98-3.02(1 \mathrm{H}, \mathrm{m}$, $\mathrm{H}-6), 3.73\left(2 \mathrm{H}, \mathrm{q}, J 7.1, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 3.83-3.86\left(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-4_{\text {endo }}\right)$, 3.89-3.93 (1H, m, H-5), 4.28-4.31 ( $\left.1 \mathrm{H}, \mathrm{m}, \mathrm{H}-\mathrm{4}_{\text {exo }}\right), 6.48(1 \mathrm{H}, \mathrm{s}$, H-2), 7.14-7.18 (3H, m, $\operatorname{Ar} H), 7.29(2 \mathrm{H}, \mathrm{d}, J 8.4, \operatorname{Ar} H), 7.40$ (2H, d, J 7.2, $\operatorname{Ar} H), 7.61(2 \mathrm{H}, \mathrm{d}, J 7.4, \operatorname{Ar} H) ; \delta_{\mathrm{C}}(101 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 14.0\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $28.1\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 34.8\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right) \text {, }}\right.$ $47.2(\mathrm{C}-6), 62.0\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 62.6(\mathrm{C}-5), 68.3(\mathrm{C}-7), 72.3(\mathrm{C}-4)$,
$81.8\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right)}\right.$, $86.2(\mathrm{C}-2), 125.0,125.9,126.0,128.3$ and 128.9 $(\mathrm{ArCH}), 136.1,137.8$ and $139.1(\mathrm{ArC}), 168.6,169.7$ and 170.3 (CO); m/z (CI, NH ${ }_{3}$ ) $534\left(\mathrm{MH}^{+}, 12 \%\right), 478$ (100); HRMS 534.2097. $\mathrm{C}_{28} \mathrm{H}_{33} \mathrm{NO}_{6}\left(\mathrm{M}+\mathrm{H}^{+}\right)$requires 534.2103.

## ( $2 R, 5 S, 6 S, 7 R$ )- and ( $2 R, 5 S, 6 S, 7 S$ )-7-p-Bromophenyl-6-tert-butoxycarbonylmethyl-7-ethoxycarbonyl-3-oxa-8-oxo-2-phenyl-1-azabicyclo[3.3.0]octanes 5 f and $\mathbf{6 f}$

According to Method 2, lactam 4 a ( $33 \mathrm{mg}, 0.08 \mathrm{mmol}$ ) was reacted with $p$-bromophenyllead( IV ) triacetate $(92 \mathrm{mg}, 0.17$ mmol ) in pyridine ( $34 \mathrm{mg}, 0.42 \mathrm{mmol}$ ) and $\mathrm{CHCl}_{3}(2 \mathrm{ml})$ at reflux for 41 h giving, after work-up, a yellow oil containing solid material. The crude product ratio $\mathbf{5 f}$ : $\mathbf{6 f}$ was $2.3: 1$. Purification by column chromatography ( $\mathrm{Et}_{2} \mathrm{O}$-petrol (40-60), 1:2) gave the inseparable products $\mathbf{5 f}: \mathbf{6 f}(34 \mathrm{mg}, 74 \%)$.

Data for 5f. $R_{\mathrm{f}} 0.29 ; v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 2979$ (w), 1721 (s), 1492 (w), 1368 (m), 1243 (m), 1153 (m), 1079 (w), 1028 (w), 1012 (w), 757 (w), 741 (w), $699(\mathrm{w}) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.31(3 \mathrm{H}, \mathrm{t}, J$ $\left.7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 1.41\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.67(1 \mathrm{H}, \mathrm{dd}, J 16.5$ and $\left.11.0, \mathrm{CH} \mathrm{HCO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 2.38\left(1 \mathrm{H}, \mathrm{dd}, 16.5\right.$ and $\left.4.0, \mathrm{CH} H \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right)$, 3.56-3.72 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-5$ and $\mathrm{H}-6), 3.97(1 \mathrm{H}, \mathrm{dd}, J 9.0$ and 7.5 , $\left.\mathrm{H}-4_{\text {endo }}\right), 4.34\left(2 \mathrm{H}, \mathrm{q}, J 7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 4.43(1 \mathrm{H}, \mathrm{dd}, J 9.0$ and $\left.6.0, \mathrm{H}-4_{\text {exo }}\right), 6.41(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-2), 6.98-7.02(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.36$ $7.50(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.54-7.57(2 \mathrm{H}, \mathrm{m}, \operatorname{Ar} H) ; \delta_{\mathrm{C}}(50.3 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 14.0\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 28.0\left(\mathrm{C}_{\left(\mathrm{CH}_{3}\right)}\right)_{3}, 35.8\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right)$, 44.3 (C-6), $61.3(\mathrm{C}-5), 62.5\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 69.3(\mathrm{C}-7), 72.5(\mathrm{C}-4)$, $81.4\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right)}\right) 87.0(\mathrm{C}-2), 122.2(\mathrm{ArC}), 126.0,128.6,128.9$, 129.8, 131.9 ( ArCH ), 133.1 ( ArC ), $138(\mathrm{ArC}), 169.4,170.3$ and $170.5(3 \times C O) ; m / z\left(\mathrm{APCI}^{+}\right) 546\left(\mathrm{M}\left({ }^{81} \mathrm{Br}\right)+\mathrm{H}^{+}, 12 \%\right)$, $544\left(\mathrm{M}\left({ }^{79} \mathrm{Br}\right)+\mathrm{H}^{+}, 8\right), 490(100)$; HRMS ( $\left.\mathrm{CI}^{+}\right)$544.1335, $\mathrm{C}_{27} \mathrm{H}_{31}{ }^{79} \mathrm{BrNO}_{6}$ requires 544.1335 .

Data for 6f. $R_{\mathrm{f}} 0.40 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.25(3 \mathrm{H}, \mathrm{t}, J 7.0$, $\left.\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 1.45\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 2.63-2.73\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CO}_{2}-\right.$ $\left.{ }^{\mathrm{t}} \mathrm{Bu}\right), 3.02-3.09(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6), 3.89(1 \mathrm{H}, \mathrm{dd}, J 9.0$ and 7.0 $\left.\mathrm{H}-4_{\text {endo }}\right), 4.08-4.15(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5), 4.21-4.38\left(3 \mathrm{H}, \mathrm{m}, \mathrm{H}-4_{\text {exo }}\right.$ and $\left.\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 6.35(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-2), 7.28-7.57$ ( $9 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ); other data as for $\mathbf{5 f}$ above.

## ( $2 R, 5 S, 6 S, 7 R$ )- and ( $2 R, 5 S, 6 S, 7 S$ )-6-tert-Butoxycarbonyl-methyl-7-ethoxycarbonyl-7-p-methoxyphenyl-3-oxa-8-oxo-2-phenyl-1-azabicyclo[3.3.0]octanes 5 g and $\mathbf{6 g}$

According to Method 3, lactam 4a ( $57 \mathrm{mg}, 0.15 \mathrm{mmol}$ ) was reacted with $p$-methoxyphenyllead(IV) triacetate $(0.20 \mathrm{~g}, 0.40$ mmol ) to give a yellow/brown oil ( 73 mg ) which was purified using flash column chromatography ( $\mathrm{Et}_{2} \mathrm{O}$-cyclohexane, $1: 2$ ) to give the inseparable products $\mathbf{5 g}: \mathbf{6 g}$ in the ratio $2.3: 1$ as a colourless glass ( $45 \mathrm{mg}, 38 \%$ ); $R_{\mathrm{f}} 0.18$ ( $\mathrm{Et}_{2} \mathrm{O}$-petrol (40-60).

Data for 5g. $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.41\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right)$, $1.68\left(1 \mathrm{H}, \mathrm{dd}, J 16.5\right.$ and $\left.11.0, \mathrm{C}_{\mathrm{H}} \mathrm{HCO}_{2}{ }^{\mathrm{H}} \mathrm{Bu}\right), 2.41(1 \mathrm{H}, \mathrm{dd}$, $J 16.5$ and $\left.4.0, \mathrm{CHHCO}{ }_{2}{ }^{\mathrm{B}} \mathrm{Bu}\right), 3.79\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 6.41(1 \mathrm{H}, \mathrm{s}$, $\mathrm{H}-2), 6.84-6.87(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.02-7.05(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H) ; m / z$ $\left(\mathrm{APCl}^{+}\right) 496\left(\mathrm{M}+\mathrm{H}^{+}, 79 \%\right) 440$ (100).

Data for 6g. $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.41\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right)$, 2.68-2.71 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}$ ), $3.82\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 6.41(1 \mathrm{H}$, s, H-2), 7.02-7.05 (2H, m, $\operatorname{ArH}$ ), 6.93-6.96 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ); $m / z$ data as above for $\mathbf{5 g}$.
( $2 R, 5 S, 6 S, 7 S$ )- and ( $2 R, 5 S, 6 S, 7 R$ )-7-Benzyl-6-tert-butoxycarb-onylmethyl-7-ethoxycarbonyl-3-oxa-8-oxo-2-phenyl-1-azabicyclo[3.3.0]octanes 5h and 6h

To a stirred suspension of pre-washed $\mathrm{NaH}(32 \mathrm{mg}, 1.4 \mathrm{mmol})$ in THF in a nitrogen atmosphere ( 3 ml ) was added a solution of the lactam $4 \mathrm{a}(0.50 \mathrm{~g}, 1.3 \mathrm{mmol})$ in THF $(6 \mathrm{ml})$ at $0^{\circ} \mathrm{C}$. Benzyl bromide ( $0.24 \mathrm{~g}, 1.4 \mathrm{mmol}$ ) was then added to the solution which was then heated at reflux for 14 h ; the reaction was
quenched by adding $\mathrm{NH}_{4} \mathrm{Cl}$ (sat. aq.) ( 2 ml ) and then water $(5 \mathrm{ml})$ to dissolve the white precipitate. The mixture was extracted with $\mathrm{DCM}(3 \times 10 \mathrm{ml})$, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated in vacuo to give a colourless oil which was purified using flash column chromatography (EtOAc-cyclohexane, 5:1) to give the products as single diastereomers as colourless oils in a total yield of $75 \%$.

Data for 5 h. $0.21 \mathrm{~g}, 34 \% ; R_{\mathrm{f}} 0.19$ (EtOAc-cyclohexane, $5: 1$ ); $[a]_{\mathrm{D}}^{23}+66.6\left(c 0.35 \mathrm{in}^{2} \mathrm{CHCl}_{3}\right.$ ); $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 2980(\mathrm{w}), 1727$ (s), $1368(\mathrm{~m}), 1221(\mathrm{~m}), 1156(\mathrm{~m}), 700(\mathrm{~m}) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $1.34\left(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 1.47\left(1 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 2.58-2.70$ ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}$ ), $2.98(1 \mathrm{H}, \mathrm{d}, J 14.0, \mathrm{C} H \mathrm{HPh}), 3.19-3.27$ ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6$ ), $3.32-3.39(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5), 3.49$ ( $1 \mathrm{H}, \mathrm{d}, J 14.0$, $\mathrm{CH} H \mathrm{Ph}), 3.78\left(1 \mathrm{H}, \mathrm{dd}, J 9.0\right.$ and $\left.7.5, \mathrm{H}-4_{\text {endo }}\right), 4.23(1 \mathrm{H}$, dd, $J 9.0$ and $\left.6.0, \mathrm{H}-4_{\text {exo }}\right), 4.25-4.34\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 6.29(1 \mathrm{H}, \mathrm{s}$, $\mathrm{H}-2), 7.05-7.36(10 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(50.3 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 14.1$ $\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 28.0\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 33.8$ and $35.7\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right.$ and $\left.\mathrm{CH}_{2} \mathrm{Ph}\right), 45.4(\mathrm{C}-6), 61.8(\mathrm{C}-5), 62.1\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 64.8(\mathrm{C}-7), 72.5$ (C-4), $81.6\left(\mathrm{C}_{( }\left(\mathrm{CH}_{3}\right)_{3}\right), 86.5(\mathrm{C}-2), 126.3,127.0,128.2,128.3$, 128.7 and $130.5(\mathrm{ArCH}), 135.5(\mathrm{ArC}), 137.8(\mathrm{ArC}), 170.6$, $171.9(3 \times C O) ; m / z\left(\mathrm{APCI}^{+}\right) 480\left(\mathrm{M}+\mathrm{H}^{+}, 58 \%\right) ;$ HRMS $\left(\mathrm{CI}^{+}\right) 480.2386, \mathrm{C}_{28} \mathrm{H}_{34} \mathrm{NO}_{6}\left(\mathrm{M}+\mathrm{H}^{+}\right)$requires 480.2386 .

Data for 6h. $0.25 \mathrm{~g}, 41 \% ; R_{\mathrm{f}} 0.29$ (EtOAc-cyclohexane, $5: 1$ ); $[a]_{\mathrm{D}}^{26}+146.3\left(c 0.23\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; v_{\max }$ (film) $/ \mathrm{cm}^{-1} 2980(\mathrm{w}), 1728(\mathrm{~s})$, $1710(\mathrm{~s}), 1368(\mathrm{~m}), 1226(\mathrm{~m}), 1155(\mathrm{~m}), 701(\mathrm{~m}) ; \delta_{\mathrm{H}}(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 1.30\left(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 1.41\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right)$, 2.26-2.42 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{C}} \mathrm{Bu}$ ), $2.55-2.62(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6), 3.12$ ( $1 \mathrm{H}, \mathrm{t}, J 8.5, \mathrm{H}-4_{\text {endo }}$ ), 3.20 ( $1 \mathrm{H}, \mathrm{d}, J 14.0, \mathrm{C} H \mathrm{HPh}$ ), $3.47(1 \mathrm{H}, \mathrm{d}$, $J 14.0, \mathrm{CH} H \mathrm{Ph}), 3.90-3.97(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5), 4.16$ ( $1 \mathrm{H}, \mathrm{dd}, J 8.5$ and $\left.6.0, \mathrm{H}-4_{\text {exo }}\right), 4.21-4.31\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 6.17(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-2)$, 7.26-7.45 ( $10 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H$ ); $\delta_{\mathrm{C}}\left(50.3 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 14.1$ $\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 28.0\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 35.13$ and $36.47\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right.$ and $\left.\mathrm{CH}_{2} \mathrm{Ph}\right), 41.2(\mathrm{C}-6), 62.0\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 62.8(\mathrm{C}-5), 65.4(\mathrm{C}-7), 72.4$ $\left.(\mathrm{C}-4), 81.4\left(\mathrm{C}-\mathrm{CH}_{3}\right)_{3}\right), 86.2(\mathrm{C}-2), 126.00,127.2,128.4,128.6$, 128.7 and $130.8(\mathrm{ArCH}), 136.0(\mathrm{ArC}), 138.1$ (ArC), 169.7, 170.6 and $171.0(3 \times C O) ; m / z\left(\mathrm{APCI}^{+}\right) 480\left(\mathrm{M}+\mathrm{H}^{+}, 16 \%\right)$, 424 (100); HRMS ( $\mathrm{CI}^{+}$) 480.2386, $\mathrm{C}_{28} \mathrm{H}_{34} \mathrm{NO}_{6}\left(\mathrm{M}+\mathrm{H}^{+}\right)$ requires 480.2386 .

## $N$-Benzylvalerolactam 8a ${ }^{22,23}$

To a stirred solution of $\delta$-valerolactam ( $1.95 \mathrm{~g}, 20 \mathrm{mmol}$ ) in dry THF ( 27 ml ) at $0{ }^{\circ} \mathrm{C}$ was added pre-washed sodium hydride $(0.57 \mathrm{~g}, 24 \mathrm{mmol})$. After stirring for 30 mins , benzyl bromide ( $3.7 \mathrm{~g}, 22 \mathrm{mmol}$ ) was added and stirring was continued at rt for 2 h and then heated under reflux for 19 h . After cooling to rt , the reaction was quenched with sat. $\mathrm{NH}_{4} \mathrm{Cl}(10 \mathrm{ml})$ and the aqueous phase washed with EtOAc $(3 \times 15 \mathrm{ml})$. The organic layers were combined, washed with water ( 15 ml ), dried $\left(\mathrm{MgSO}_{4}\right)$ and the solvent removed in vacuo to yield a yellow oil. The oil was purified by flash column chromatography (petrolEtOAc, 2:1) to give the title compound ( $1.6 \mathrm{~g}, 85 \%$ ). $R_{\mathrm{f}} 0.11$ (petrol-EtOAc, 2:1); $v_{\max }($ film $) / \mathrm{cm}^{-1} 3028(\mathrm{w}), 2945(\mathrm{~m}), 2965$ (m), $1650(\mathrm{~s}) ; \delta_{\mathrm{H}}\left(200 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.78-1.81(4 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{H}-4$ and $2 \times \mathrm{H}-5), 2.48(2 \mathrm{H}, \mathrm{t}, J 6.5,2 \times \mathrm{H}-3), 3.20(2 \mathrm{H}, \mathrm{t}, J 6.0$, $2 \times \mathrm{H}-6), 4.61\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{Ph}\right), 7.25-7.35(5 \mathrm{H}, \mathrm{m}, \mathrm{ArH}) ; m / z$ $\left(\mathrm{CI}, \mathrm{NH}_{3}\right) 190\left(\mathrm{MH}^{+}, 100 \%\right)$.

## $N$-Benzyl-3-chlorovalerolactam 8b and $N$-benzyl-3,3-dichlorovalerolactam

To a stirred solution of lactam $\mathbf{8 b}(1.3 \mathrm{~g}, 7.0 \mathrm{mmol})$ in dry THF ( 10 ml ) at $-78^{\circ} \mathrm{C}$ under an inert atmosphere, was added secbutyllithium ( 0.91 M in hexanes, $8.5 \mathrm{ml}, 7.7 \mathrm{mmol}$ ) and the reaction mixture was stirred for 30 mins . Toluene- $p$-sulfonyl chloride ( $2.0 \mathrm{~g}, 11 \mathrm{mmol}$ ) in dry THF ( 5 ml ) was added and the reaction mixture slowly warmed to rt. After 16 h the resulting suspension was quenched with sat. $\mathrm{NH}_{4} \mathrm{Cl}(20 \mathrm{ml})$ and the aqueous layer was washed with $\operatorname{EtOAc}(3 \times 10 \mathrm{ml})$. The organic
layers were combined, washed with water ( 15 ml ), dried $\left(\mathrm{MgSO}_{4}\right)$, and the solvent removed in vacuo to yield a yellow oil. Purification by flash column chromatography (petrol-EtOAc, $3: 2$ ) afforded both the mono-substituted product $\mathbf{8 b}$ as a yellow solid ( $1.11 \mathrm{~g}, 71 \%$ ) and the dichloro product as an orange solid ( $0.075 \mathrm{~g}, 4 \%$ ).

Data for 8b. $R_{\mathrm{f}} 0.28$ (petrol-EtOAc, 2:1); mp $72-73^{\circ} \mathrm{C}$; $v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3012(\mathrm{~m}), 2963(\mathrm{~m}), 1649(\mathrm{~s}), 1494(\mathrm{~m}) ; \delta_{\mathrm{H}}$ ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $1.79(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5), 2.22(3 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{H}-4$ and $\mathrm{H}-5), 3.25(2 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{H}-6), 4.46(1 \mathrm{H}, \mathrm{d}, J 14.6, \mathrm{C} H \mathrm{HPh}), 4.52$ $(1 \mathrm{H}, \mathrm{t}, J 4.5, \mathrm{C} H \mathrm{Cl}), 4.73(1 \mathrm{H}, \mathrm{d}, J 14.6, \mathrm{CH} H \mathrm{Ph}), 7.25-7.36$ $(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 18.41(\mathrm{C}-5), 30.95(\mathrm{C}-4)$, 46.89 (C-6), 50.46 ( $\mathrm{CH}_{2} \mathrm{Ph}$ ), 54.97 (C-3), 127.55, 127.97 and $128.62(\mathrm{ArCH}), 136.38,(\mathrm{ArC}), 166.02(\mathrm{C}-2) ; m / z\left(\mathrm{CI}, \mathrm{NH}_{3}\right) 226$ $\left(\mathrm{MH}^{+},{ }^{37} \mathrm{Cl}, 12 \%\right), 224\left(\mathrm{MH}^{+},{ }^{35} \mathrm{Cl}, 39\right), 190$ (100); HRMS 224.0842. $\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{NOCl}\left(\mathrm{M}+\mathrm{H}^{+}\right)$requires 224.0842 .

Dichlorovalerolactam. $R_{\mathrm{f}} 0.47$ (petrol-EtOAc, 2:1); mp 70$73{ }^{\circ} \mathrm{C} ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 2969(\mathrm{~m}), 1678(\mathrm{~s}) ; \delta_{\mathrm{H}}(500 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 2.01-2.07(2 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{H}-5), 2.77-2.80(2 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{H}-$ 4), $3.30(2 \mathrm{H}, \mathrm{t}, J 6.2,2 \times \mathrm{H}-6), 4.60\left(2 \mathrm{H}, \mathrm{s}, \mathrm{C}_{2} \mathrm{Ph}\right), 7.25-7.36$ $(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 19.93(\mathrm{C}-5), 43.56(\mathrm{C}-4)$, 47.13 (C-6), $51.30\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 82.92(\mathrm{C}-3), 127.73,127.93$ and $128.71(\mathrm{ArCH}), 135.84(\mathrm{ArC}), 163.68(\mathrm{C}-2) ; m / z\left(\mathrm{CI}, \mathrm{NH}_{3}\right) 262$ $\left(\mathrm{M}+\mathrm{H}^{+}, 12 \%\right), 260(69), 258$ (100), 224 (35), 190 (100).

## $N$-Benzyl-3-iodovalerolactam 8c

A solution of lactam $\mathbf{8 b}(388 \mathrm{mg}, 1.2 \mathrm{mmol})$ and sodium iodide ( $32 \mathrm{mg}, 2.2 \mathrm{mmol}$ ) in acetone ( 10 ml ) were heated under reflux for 17 h . After cooling to rt, the resulting precipitate was removed by filtration and the filtrate concentrated in vacuo. The concentrate was dissolved in EtOAc ( 10 ml ), washed with water ( 10 ml ), dried $\left(\mathrm{MgSO}_{4}\right)$ and the solvent removed in vacuo. Purification by flash column chromatography (petrol-EtOAc, 1:1) afforded the title compound $\mathbf{8 c}$ as a yellow oil ( $510 \mathrm{mg}, 90 \%$ ). $R_{\mathrm{f}}$ 0.40 (petrol-EtOAc, 1:1); $v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3007(\mathrm{~m}), 1636(\mathrm{~s}) ;$ $\delta_{\mathrm{H}}\left(200 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.78-1.85(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5), 2.01-2.10(1 \mathrm{H}$, $\mathrm{m}, \mathrm{H}-5), 2.18-2.29(2 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{H}-4), 3.35-3.42(2 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{H}-$ $6), 4.42$ ( $1 \mathrm{H}, \mathrm{d}, J 14.5, \mathrm{C} H \mathrm{HPh}), 4.82$, ( $1 \mathrm{H}, \mathrm{d}, J 14.5, \mathrm{CH} H \mathrm{Ph}$ ), 4.92-4.95 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CHCl})$, 7.26-7.37 ( $5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H)$; $\delta_{\mathrm{C}}(125.8$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 20.53(\mathrm{C}-5), 23.17$ (C(3)), 32.49 (C-4), 46.85 (C-6), $50.52\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 127.52,128.00$ and $128.66(\mathrm{ArCH})$, $136.65(\mathrm{ArC}), 167.84(\mathrm{C}-2) ; m / z\left(\mathrm{CI}, \mathrm{NH}_{3}\right) 316\left(\mathrm{MH}^{+}, 42 \%\right)$, 190 (100); HRMS 316.0198, $\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{NOI}\left(\mathrm{M}+\mathrm{H}^{+}\right)$requires 316.1476 .

## ( $2 R, 5 S, 6 S, 7 R$ )- and ( $2 R, 5 S, 6 S, 7 S$ )-7-( $N$-Benzyl-2-oxopiperidin-3-yl)-6-tert-butoxycarbonylmethyl-7-ethoxycarbonyl-3-oxa-8-oxo-2-phenyl-1-azabicyclo[3.3.0]octanes 9a and 9b

Lactam $\mathbf{4 a}(195 \mathrm{mg}, 0.50 \mathrm{mmol})$ in dry THF ( 2 ml ) was reacted with sodium hydride ( $17 \mathrm{mg}, 0.70 \mathrm{mmol}$ ) and subsequently with a solution of iodolactam $\mathbf{8 c}(150 \mathrm{mg}, 0.48 \mathrm{mmol})$ in dry THF $(2 \mathrm{ml})$. Purification by flash column chromatography (petrolEtOAc, 3:2), gave the two separable diastereomers $9 \mathbf{a}$ and $9 \mathbf{b}$ in a 1.1:1 ratio and as pale yellow oils ( $102 \mathrm{mg}, 36 \%$ ).

Data for 9a. $54 \mathrm{mg}, 19 \% ; R_{\mathrm{f}} 0.64$ (petrol-EtOAc, 1:1); $v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3020(\mathrm{~m}), 1718(\mathrm{~s}), 1637(\mathrm{~s}) ; \delta_{\mathrm{H}}(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 1.27\left(3 \mathrm{H}, \mathrm{t}, J 7.2, \mathrm{CH}_{2} \mathrm{CH}_{2}\right), 1.44\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right)$, 1.69-1.96, ( $4 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{CH}_{2}$ ), 2.06-2.09 ( $1 \mathrm{H}, \mathrm{m}$, ring proton), 2.88-2.92 ( $1 \mathrm{H}, \mathrm{m}$, ring proton), 3.18-3.27 ( $3 \mathrm{H}, \mathrm{m}, \mathrm{H}-6$ and $\left.\mathrm{CH}_{2}\right), 3.44-3.48(1 \mathrm{H}, \mathrm{m}$, ring proton), 3.86-3.92 $(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-$ $\left.4_{\text {endo }}\right), 4.06-4.15(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-5), 4.21-4.31\left(3 \mathrm{H}, \mathrm{m}, \mathrm{H}-4_{\text {exo }}\right.$ and $\left.\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.53(1 \mathrm{H}, \mathrm{d}, J 14.8, \mathrm{C} H \mathrm{HPh}), 4.58(1 \mathrm{H}, \mathrm{d}, J 14.8$, $\mathrm{CH} H \mathrm{Ph}), 6.27(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-2), 7.16-7.47(10 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H) ; \delta_{\mathrm{C}}(100.7$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 13.96\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 22.69\left(\mathrm{CH}_{2}\right), 22.99\left(\mathrm{CH}_{2}\right)$, $\left.28.09\left(\mathrm{C}_{\left(\mathrm{CH}_{3}\right)}\right)_{3}\right), 35.11\left(\mathrm{CH}_{2}\right), 42.00(\mathrm{C}-6), 45.02\left(\mathrm{CH}_{2}\right), 48.82$ $\left(\mathrm{CH}_{2}\right), 50.29\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 61.47\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 62.81(\mathrm{C}-5), 65.60$
(C-7), $72.47(\mathrm{C}-4), 80.77\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 86.68(\mathrm{C}-2), 126.11,127.30 \text {, }}\right.$ 127.40, 127.85, 128.14, 128.40 and $128.50(\mathrm{ArCH}), 136.87$, 137.01 and $138.06(\mathrm{ArC}), 168.99,170.03,171.72$ and 172.17 (CO); m/z (CI, NH $\mathrm{N}_{3}$ ) $577\left(\mathrm{MH}^{+}, 100 \%\right)$, 521 (8); HRMS 577.2907. $\mathrm{C}_{33} \mathrm{H}_{41} \mathrm{~N}_{2} \mathrm{O}_{7}\left(\mathrm{MH}^{+}\right)$requires 577.2914.

Data for 9b. $48 \mathrm{mg}, 17 \% ; R_{\mathrm{f}} 0.54$ (petrol-EtOAc, 1:1); $v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3020(\mathrm{~m}), 1718(\mathrm{~s}), 1637(\mathrm{~s}) ; \delta_{\mathrm{H}}(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 1.32\left(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.46\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right)$, $1.50-1.56,\left(1 \mathrm{H}, \mathrm{m}\right.$, ring proton), $1.82-1.96\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}\right), 2.21-$ $2.29\left(1 \mathrm{H}, \mathrm{m}\right.$, ring proton), $2.53\left(2 \mathrm{H}, \mathrm{d}, J 5.4, \mathrm{CH}_{2}\right), 2.84-2.89$ ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6$ ), $3.16-3.22\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}\right), 3.36(1 \mathrm{H}, \mathrm{dd}, J 12.8$ and 5.1, ring proton), 3.89-3.91 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-4_{\text {endo }}$ and H-5), 4.26-4.37 $\left(3 \mathrm{H}, \mathrm{m}, \mathrm{H}-4_{\text {exo }}\right.$ and $\left.\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 4.60(1 \mathrm{H}, \mathrm{d}, J 14.8, \mathrm{C} H \mathrm{HPh})$, $4.63(1 \mathrm{H}, \mathrm{d}, J 14.8, \mathrm{CH} H \mathrm{Ph}), 6.37(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-2), 7.25-7.50(10 \mathrm{H}$, $\mathrm{m}, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(100.7 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 14.12\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 22.63$ $\left(\mathrm{CH}_{2}\right), 23.41\left(\mathrm{CH}_{2}\right), 28.02\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 36.44\left(\mathrm{CH}_{2}\right), 41.77(\mathrm{C}-6) \text {, }}\right.$ $44.83\left(\mathrm{CH}_{2}\right), 46.06\left(\mathrm{CH}_{2}\right), 50.52\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 61.74\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, 63.14 (C-5), $65.56(\mathrm{C}-7), 71.78(\mathrm{C}-4), 81.44\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 8} 87.08\right.$ (C-2), 126.02, 126.26, 127.18, 127.51, 128.00, 128.04, 128.31 and $128.65(\mathrm{ArCH}), 137.15$ and $138.36(\mathrm{ArC}), 167.85,169.78$, 171.01 and $171.78(\mathrm{CO}) ; m / z\left(\mathrm{CI}, \mathrm{NH}_{3}\right) 577\left(\mathrm{MH}^{+}, 100 \%\right), 521$ (8); HRMS 577.2907, $\mathrm{C}_{33} \mathrm{H}_{41} \mathrm{~N}_{2} \mathrm{O}_{7}\left(\mathrm{MH}^{+}\right)$requires 577.2914.

## General method for deprotection

To a solution of the $N, O$-acetal in DCM at rt was added TFA. After the specified time period, the acid was neutralised by careful addition of saturated aqueous sodium hydrogen carbonate solution or an aqueous solution of sodium hydroxide with stirring in an ice bath. After separation of the two layers, and extraction of the aqueous with DCM, the organics were dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated in vacuo to give the crude product which was purified using flash column chromatography.

## (2S,3S,4R)- and ( $2 S, 3 S, 4 S$ )-3-tert-Butoxycarbonylmethyl-4-ethoxycarbonyl-2-hydroxymethyl-5-oxo-4-phenylpyrrolidines 10a and 11a

Following the general method, the lactam 5a, $\mathbf{6 a}(2.1: 1)(0.17 \mathrm{~g}$, 0.35 mmol ), was reacted with TFA ( 1.5 ml ) in DCM ( 10 ml ) for 45 min . Purification using flash column chromatography (EtOAc-petrol (40-60), 1:1 and then EtOAc-petrol (40-60)$\mathrm{MeOH}, 200: 200: 15)$ gave the title compound as two diastereomers in a combined yield of $81 \%$.

Data for 10a. White crystalline solid ( $70 \mathrm{mg}, 53 \%$ ); $R_{\mathrm{f}} 0.24$ (EtOAc-petrol (40-60)-MeOH, 20:15:2); [a $]_{\mathrm{D}}^{22}-42.0$ (c 0.35 , $\left.\mathrm{CHCl}_{3}\right) ; \mathrm{mp} 138-142^{\circ} \mathrm{C} ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3424(\mathrm{~m}), 3248(\mathrm{~m}$, br), 1720 (s), 1369 (m), 1228 (s), 1154 (s), 1095 (m), 1055 (m), 1037 (m), $946(\mathrm{w}), 701(\mathrm{~m}) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$, c 34.6 mg $\left.\mathrm{ml}^{-1}\right) 1.30\left(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 1.40\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.86$ ( $2 \mathrm{H}, \mathrm{d}, J 7.5, \mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}$ ), $3.24-3.29(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3), 3.55(1 \mathrm{H}$, br s, OH), 3.53-3.61 (3H, m, H-2 and CHHOH), $3.76(1 \mathrm{H}, \mathrm{dd}$, $J 11.5$ and $3.0, \mathrm{CH} H \mathrm{OH}), 4.24-4.41\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2}\right)$, $7.19-$ $7.38(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.56(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{N} H) ; \delta_{\mathrm{C}}(125.8 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 13.96\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right)$, $27.95\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 36.01\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right)$, $41.07(\mathrm{C}-3), 59.01(\mathrm{C}-2), 62.42\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 63.63\left(\mathrm{CH}_{2} \mathrm{OH}\right)$, $64.85(\mathrm{C}-4), 81.22\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 127.83,128.45 \text { and } 128.61}\right.$ $(\mathrm{ArCH}), 134.13(\mathrm{C}-6), 170.47,171.31$ and $173.87(3 \times \mathrm{CO}) ; m / z$ ( $\mathrm{APCI}^{+}$) $378\left(\mathrm{M}+\mathrm{H}^{+}, 14 \%\right), 322$ (100), 250 (9); HRMS 378.1917, $\mathrm{C}_{20} \mathrm{H}_{28} \mathrm{NO}_{6}$ requires 378.1917.

Data for 11a. Colourless oil ( $38 \mathrm{mg}, 28 \%$ ); $R_{\mathrm{f}} 0.32$ (EtOAcpetrol ( $40-60$ )-MeOH, $20: 15: 2$ ); $[a]_{\mathrm{D}}^{23}-14.1\left(c 0.425\right.$ in $\mathrm{CHCl}_{3}$ ); $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 3336(\mathrm{~m}, \mathrm{br}), 2979(\mathrm{~m}), 1724(\mathrm{~s}), 1368(\mathrm{~m}), 1300$ (m), $1232(\mathrm{~m}), 1153(\mathrm{~s}), 1063(\mathrm{~m}), 847(\mathrm{w}), 699(\mathrm{~m}) ; \delta_{\mathrm{H}}(300$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}, c 9 \mathrm{mg} \mathrm{ml}^{-1}\right) 1.28\left(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 1.44$ $\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 2.53\left(1 \mathrm{H}, \mathrm{dd}, J 17.0\right.$ and $\left.10.0, \mathrm{CH} \mathrm{HCO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right)$, $2.72\left(2 \mathrm{H}, \mathrm{dd}, J 17.0\right.$ and $2.5, \mathrm{CH} \mathrm{COO}_{2}{ }^{ } \mathrm{Bu}$ and OH$), 3.18-3.25$ $(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3), 3.52(1 \mathrm{H}, \mathrm{dd}, J 12.0$ and $5.5, \mathrm{C} H \mathrm{HOH}), 3.67-$
$3.71(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-2), 3.80(1 \mathrm{H}, \mathrm{dd}, J 12.0$ and $3.0, \mathrm{CHHOH})$, 4.22-4.32 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2}$ ), 6.41 ( $1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{NH}$ ), 7.29-7.41 $(3 \mathrm{H}, \mathrm{m}, \operatorname{ArH}), 7.44-7.48\left(2 \mathrm{H}, \mathrm{m}, \mathrm{ArH}\right.$ ortho- to C-6); $\delta_{\mathrm{C}}$ $\left(50.3 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 14.1\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 27.9\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 34.6}\right.$ $\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 42.4(\mathrm{C}-3), 59.9(\mathrm{C}-2), 62.0\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 62.8$ $\left(\mathrm{CH}_{2} \mathrm{OH}\right), 63.8(\mathrm{C}-4), 81.8\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right)}\right), 127.7,128.0$ and 128.3 $(\mathrm{ArCH}), 136.3(\mathrm{C}-6), 169.6,171.8$ and $173.1(3 \times \mathrm{CO}) ; \mathrm{m} / \mathrm{z}$ $\left(\mathrm{APCI}^{+}\right) 378\left(\mathrm{M}+\mathrm{H}^{+}, 8 \%\right), 322(100)$.

## ( $2 S, 3 S, 4 R$ )- and ( $2 S, 3 S, 4 S$ )-3-tert-Butoxycarbonylmethyl-4-ethoxycarbonyl-2-hydroxymethyl-4-p-methoxyphenyl-5-oxopyrrolidines 10b and 11b

Following the general method, lactam $\mathbf{5 g}, \mathbf{6 g}(31 \mathrm{mg})$, was reacted with TFA ( 0.3 ml ) in DCM ( 2 ml ) for 45 min . Purification by recrystallisation from EtOAc-petrol (40-60) gave the major diastereomer 10b as a white solid ( $4 \mathrm{mg}, 25 \%$ ). Purification of the mother liquor by chromatography (EtOAc-petrol $(40-60)-\mathrm{MeOH}, 20: 20: 2$ ) gave the minor diastereomer 11b as a colourless oil ( $2 \mathrm{mg}, 13 \%$ ).

Data for 10b. White crystalline solid; $R_{\mathrm{f}} 0.22$ (EtOAc-petrol (40-60)-MeOH, $20: 20: 2$ ); mp 156- $159{ }^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}^{25}-153.3(c 0.015$ in $\left.\mathrm{CHCl}_{3}\right) ; v_{\max }\left(\mathrm{CDCl}_{3}, c 7.5 \mathrm{mg} \mathrm{ml}^{-1}\right) / \mathrm{cm}^{-1} 3425(\mathrm{~m}), 3330(\mathrm{~m}$, br), 2983 (w), 1718 (s), 1613 (w), 1514 (m), 1477 (m), 1370 (m), 1253 (s), 1154 (s); $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}, c 7.3 \mathrm{mg} \mathrm{ml}^{-1}\right) 1.30$ $\left(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 1.41\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.85(2 \mathrm{H}, \mathrm{dd}$, $J 17.0$ and $\left.9.0, \mathrm{CHHCO}_{2}{ }^{\mathrm{E}} \mathrm{Bu}\right), 1.96(2 \mathrm{H}, \mathrm{dd}, J 17.0$ and 6.0 , $\left.\mathrm{CH} H \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 2.74(1 \mathrm{H}$, br $\mathrm{t}, J 6.0, \mathrm{OH}), 3.27-3.33(1 \mathrm{H}, \mathrm{m}$, $\mathrm{H}-3), 3.51-3.65(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-2$ and CHHOH$), 3.78-3.88(4 \mathrm{H}, \mathrm{m}$, $\mathrm{CH} H \mathrm{OH}$ and $\left.\mathrm{OCH}_{3}\right), 4.26-4.38\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 6.83(1 \mathrm{H}$, br s, $\mathrm{N} H), 6.86-6.91(2 \mathrm{H}, \mathrm{m}, \operatorname{Ar} H), 7.12-7.17(2 \mathrm{H}, \mathrm{m}, \operatorname{Ar} H) ; \delta_{\mathrm{C}}$ $\left(125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$, c $\left.7.5 \mathrm{mg} \mathrm{ml}^{-1}\right) 13.98\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 27.98$ $\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 36.27\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 41.11(\mathrm{C}-3), 55.29\left(\mathrm{OCH}_{3}\right)$, $58.72(\mathrm{C}-2), 62.42\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 63.89(\mathrm{C}-4), 64.00\left(\mathrm{CH}_{2} \mathrm{OH}\right)$, $81.35\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 114.04(\mathrm{ArC}), 126.22(\mathrm{C}-6), 129.57(\mathrm{ArC}) \text {, }}\right.$ $158.99(\mathrm{MeO}-\mathrm{C}), 170.63,171.50$ and $173.39(3 \times \mathrm{CO}) ; \mathrm{m} / \mathrm{z}$ $\left(\mathrm{APCI}^{+}\right) 408\left(\mathrm{M}+\mathrm{H}^{+}, 25 \%\right), 352(100), 280(23)$.

Data for 11b. $R_{\mathrm{f}} 0.27$ (EtOAc-petrol (40-60)-MeOH, $20: 20: 2$ ); $[a]_{\mathrm{D}}^{26}-18.0$ (c 0.05 in $\mathrm{CHCl}_{3}$ ); $v_{\text {max }}$ (film) $/ \mathrm{cm}^{-1} 3351$ (m), 1725 (s), 1712 (s), 1515 (m), 1298 (m), 1254 (s), 1183 (m), $1155(\mathrm{~s}), 1068(\mathrm{~m}), 1037(\mathrm{~m}) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}, c 2 \mathrm{mg} \mathrm{ml}^{-1}\right)$ $1.29\left(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 1.45\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 2.52(1 \mathrm{H}$, dd, $J 17.0$ and $\left.10.0, \mathrm{C} H \mathrm{HCO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 2.71(1 \mathrm{H}, \mathrm{dd}, J 17.0$ and 2.5 , $\left.\mathrm{CH} H \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 3.17-3.21(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3), 3.55(1 \mathrm{H}, \mathrm{dd}, J 12.0$ and 5.5, $\mathrm{C} H \mathrm{HOH}), 3.66-3.70(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-2), 3.80-3.83(4 \mathrm{H}, \mathrm{m}$, CHHOH and $\left.\mathrm{OCH}_{3}\right), 4.23-4.29\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 6.06(1 \mathrm{H}$, $\mathrm{br} \mathrm{s}, \mathrm{N} H), 6.90-6.93(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.38-7.41(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H)$; $\delta_{\mathrm{C}}\left(125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}, c 2 \mathrm{mg} \mathrm{ml}{ }^{-1}\right) 14.15\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 27.98$ $\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 34.47\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{B}} \mathrm{Bu}\right), 42.46(\mathrm{C}-3), 55.26\left(\mathrm{OCH}_{3}\right)$, $59.57(\mathrm{C}-2), 62.03\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 62.93\left(\mathrm{C}-4\right.$ and $\left.\mathrm{CH}_{2} \mathrm{OH}\right), 81.98$ $\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 113.80(\mathrm{ArC} \text { meta- to C-6), } 128.26(\mathrm{C}-6), 129.13}\right.$ $(\mathrm{ArC}), 159.06(\mathrm{MeOC}), 169.86,172.21$ and $172.87(3 \times \mathrm{CO})$; $m / z\left(\mathrm{APCI}^{+}\right) 408\left(\mathrm{M}+\mathrm{H}^{+}, 23 \%\right), 352(100)$.

## (2S,3S,4S)-4-Benzyl-3-tert-butoxycarbonylmethyl-4-ethoxy-carbonyl-2-hydroxymethyl-5-oxopyrrolidine 10c

Following the general method, the lactam 5h ( $85 \mathrm{mg}, 0.22$ mmol ), was reacted with TFA ( 0.5 ml ) in DCM ( 10 ml ) for 1 h . Purification using flash column chromatography (EtOAc-petrol (40-60), 1:1 then EtOAc-petrol (40-60)-MeOH, 20:15:1) gave the title compound as a colourless oil, as a single diastereomer ( $60 \mathrm{mg}, 86 \%$ ); $R_{\mathrm{f}} 0.20$ (EtOAc-petrol ( $40-60$ )- MeOH , 20:15:1); $[a]_{\mathrm{D}}^{23}+13.91\left(c 0.115\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; v_{\max }\left(\mathrm{CDCl}_{3}, c 9 \mathrm{mg}\right.$ $\left.\mathrm{ml}^{-1}\right) / \mathrm{cm}^{-1} 3424(\mathrm{~m}), 3245(\mathrm{~m}, \mathrm{br}), 1719(\mathrm{~s}), 1370(\mathrm{~m}), 1251(\mathrm{~m})$, $1155(\mathrm{~s}), 1098(\mathrm{~s}), 1040(\mathrm{~m}) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}, c 9 \mathrm{mg} \mathrm{ml}^{-1}\right)$ $1.29\left(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 1.48\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 2.42(1 \mathrm{H}$, dd, $J 16.5$ and $\left.7.5, \mathrm{CHHCO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 2.59(1 \mathrm{H}, \mathrm{br} \mathrm{t}, \mathrm{OH}), 2.70-$ $2.76(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3), 2.76\left(1 \mathrm{H}, \mathrm{dd}, J 16.5\right.$ and $\left.7.0, \mathrm{CH} H \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right)$,
$3.06(1 \mathrm{H}, \mathrm{d}, J 14.5, \mathrm{C} H \mathrm{HPh}), 3.24(1 \mathrm{H}, \mathrm{dd}, J 16.0$ and $7.5, \mathrm{H}-$ 2), $3.40-3.47(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{HOH}), 3.45(1 \mathrm{H}, \mathrm{d}, J 14.5, \mathrm{CH} H \mathrm{Ph})$, 3.60-3.69 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CHHOH}$ ), 4.20-4.31 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2}$ ), 6.43 $(1 \mathrm{H}, \mathrm{s}, \mathrm{N} H), 7.23-7.34(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(50.3 \mathrm{MHz}, \mathrm{CDCl}_{3}, c\right.$ $\left.64 \mathrm{mg} \mathrm{ml}^{-1}\right) 13.9\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right)$, $28.0\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right)$, 33.4 and 35.2 $\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right.$ and $\left.\mathrm{CH}_{2} \mathrm{Ph}\right), 41.3(\mathrm{C}-3), 59.3(\mathrm{C}-2), 59.8(\mathrm{C}-4)$, $61.9\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right)$, $63.1\left(\mathrm{CH}_{2} \mathrm{OH}\right), 81.5\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 127.0,128.3$ and $130.2(\mathrm{ArCH}), 135.9(\mathrm{C}-7), 170.9,171.6$ and 175.3 $(3 \times C O) ; m / z\left(\mathrm{APCI}^{+}\right) 392\left(\mathrm{M}+\mathrm{H}^{+}, 6 \%\right), 336$ (100); HRMS 392.2073, $\mathrm{C}_{21} \mathrm{H}_{30} \mathrm{NO}_{6}$ requires 392.2073.

## (2S,3S,4R)-4-Benzyl-3-tert-butoxycarbonylmethyl-4-ethoxy-carbonyl-2-hydroxymethyl-5-oxopyrrolidine 11c

Following the general method, the lactam 6h $(0.11 \mathrm{~g}, 0.23$ $\mathrm{mmol})$, was reacted with TFA $(0.65 \mathrm{ml})$ in $\operatorname{DCM}(13 \mathrm{ml})$ for 1 h . Purification using flash column chromatography (EtOAcpetrol ( $40-60$ )- $\mathrm{MeOH}, 20: 15: 1$ ) gave the title compound as a colourless crystalline solid, as a single diastereomer ( 70 mg , $77 \%$ ); $R_{\mathrm{f}} 0.27$ (EtOAc-petrol (40-60)-MeOH, 20:15:1); mp $110-111{ }^{\circ} \mathrm{C} ;[a]_{\mathrm{D}}^{24}+58.3\left(c 0.12\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; v_{\max }\left(\mathrm{CDCl}_{3}, c 10 \mathrm{mg}\right.$ $\left.\mathrm{ml}^{-1}\right) / \mathrm{cm}^{-1} 3424(\mathrm{~m}), 3333(\mathrm{~m}, \mathrm{br}), 1710(\mathrm{~s}), 1369(\mathrm{~m}), 1300(\mathrm{~m})$, $1198(\mathrm{~m}), 1153(\mathrm{~s}) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}, c 10 \mathrm{mg} \mathrm{ml}^{-1}\right) 1.33$ $\left(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 1.46\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 2.11(1 \mathrm{H}, \mathrm{t}, J 6.0$, $\mathrm{OH}), 2.29\left(1 \mathrm{H}, \mathrm{dd}, J 16.5\right.$ and $\left.8.5, \mathrm{C}^{2} \mathrm{HCO}_{2}{ }^{\mathrm{B}} \mathrm{Bu}\right), 2.38(1 \mathrm{H}, \mathrm{dd}$, $J 16.5$ and $\left.5.0, \mathrm{CHHCO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 2.60-2.67(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3), 2.98-$ $3.06(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-2), 3.15(1 \mathrm{H}, \mathrm{d}, J 14.0, \mathrm{C} H \mathrm{HPh}), 3.40(1 \mathrm{H}, \mathrm{d}$, $J$ 14.0, CHHPh $), 3.47-3.54\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}\right), 4.18-4.36(2 \mathrm{H}$, $\left.\mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 6.20(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{N} H), 7.21-7.31(5 \mathrm{H}, \mathrm{m}, \mathrm{ArCH})$; $\delta_{\mathrm{C}}\left(50.3 \mathrm{MHz}, \mathrm{CDCl}_{3}, c 98 \mathrm{mg} \mathrm{ml}{ }^{-1}\right) 14.2\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 27.9$ $\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 35.5$ and $36.5\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{H}} \mathrm{Bu}\right.$ and $\left.\mathrm{CH}_{2} \mathrm{Ph}\right), 37.3(\mathrm{C}-3)$, $60.2(\mathrm{C}-2), 60.7(\mathrm{C}-4), 61.8\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 63.9\left(\mathrm{CH}_{2} \mathrm{OH}\right), 81.5$ $\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 126.9,128.3 \text { and } 130.9(\mathrm{ArCH}), 135.9(\mathrm{C}-7), 170.1 \text {, }}^{\text {, }}\right.$ 170.0 and $174.3(3 \times C O) ; m / z\left(\mathrm{APCI}^{+}\right) 392\left(\mathrm{M}+\mathrm{H}^{+}, 21 \%\right)$, 336 (100); HRMS 392.2073, $\mathrm{C}_{21} \mathrm{H}_{30} \mathrm{NO}_{6}\left(\mathrm{M}+\mathrm{H}^{+}\right)$requires 392.2073.

## ( $2 S, 3 S, 4 R$ )- and ( $2 S, 3 S, 4 S$ )-3-tert-Butoxycarbonylmethyl-2-hydroxymethyl-5-oxo-4-phenylpyrrolidines 12 and 13

To a solution of 10a ( $69 \mathrm{mg}, 0.18 \mathrm{mmol}$ ) in $\mathrm{EtOH}(5 \mathrm{ml})$ at rt was added $\mathrm{NaOH}(1 \mathrm{M}$, aq., $1.1 \mathrm{ml}, 1.1 \mathrm{mmol})$ with stirring. After 5 h , water ( 15 ml ) was added and the solution extracted with EtOAc ( $5 \times 15 \mathrm{ml}$ ). Drying $\left(\mathrm{MgSO}_{4}\right)$ and evaporation in vacuo gave the title compound as a white solid ( $17 \mathrm{mg}, 30 \%$ ). Acidification of the aqueous layer with $\mathrm{HCl}(1 \mathrm{M}$, aq.) then extraction with EtOAc, drying $\left(\mathrm{MgSO}_{4}\right)$ and evaporation in vacuo gave a colourless gum ( 42 mg ), which was a mixture of the title compounds and the acid 14 . The combined material was heated in vacuo ( 0.8 mbar ) at $135^{\circ} \mathrm{C}$ for 30 min to give the title compounds as a white solid consisting of a $1: 1$ mixture of $\mathbf{1 2}$ and $\mathbf{1 3}(46 \mathrm{mg}, 82 \%) . R_{\mathrm{f}} 0.28$ (EtOAc-petrol (40-60)$\mathrm{MeOH}, 20: 15: 2) ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3424(\mathrm{~m}), 3312(\mathrm{~m}, \mathrm{br})$, 1698 (s), 1369 (m), 1152 (s), 1090 (m);

Data for 12. $\delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}, c 2 \mathrm{mg} \mathrm{ml}^{-1}\right) 1.38(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.87\left(1 \mathrm{H}\right.$, dd, $J 17.0$ and $\left.7.5, \mathrm{C} H \mathrm{HCO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 1.96$ $(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 2.10\left(1 \mathrm{H}, \mathrm{dd}, J 17.0\right.$ and $\left.8.0, \mathrm{CHHCO}{ }_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right)$, 2.84-2.90 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3$ ), $3.50-3.53(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-2), 3.55-3.61(1 \mathrm{H}$, $\mathrm{m}, \mathrm{C} H \mathrm{HOH}), 3.79(1 \mathrm{H}, \mathrm{dd}, J 11.0$ and $3.5, \mathrm{CHHOH}), 3.98$ ( $1 \mathrm{H}, \mathrm{d}, J 9.5, \mathrm{H}-4$ ), $6.68(1 \mathrm{H}, \mathrm{s}, \mathrm{N} H), 7.13-7.15(2 \mathrm{H}, \mathrm{m}, \operatorname{Ar} H)$, 7.24-7.38 ( $3 \mathrm{H}, \mathrm{m}, \operatorname{Ar} H) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{c} 30 \mathrm{mg} \mathrm{ml}^{-1}\right)$ $27.85\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 35.27\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 37.47(\mathrm{C}-3), 51.60(\mathrm{C}-4)$, $60.29(\mathrm{C}-2), 64.07\left(\mathrm{CH}_{2} \mathrm{OH}\right), 80.80\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 127.41,128.71 \text {, }}^{\text {, }}\right.$ 128.78 and $129.26(\mathrm{ArCH}), 135.22(\mathrm{C}-6), 171.16$ and 178.74 $(2 \times C O)$; other data as below for 13 .

Data for 13. $\delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}, c 2 \mathrm{mg} \mathrm{ml}^{-1}\right) 1.38(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.96(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 2.44-2.53\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right)$, 2.58-2.65 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3$ ), 3.46 ( $1 \mathrm{H}, \mathrm{d}, J 10.0, \mathrm{H}-4$ ), $3.55-3.61$ $(2 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{HOH}$ and $\mathrm{H}-2), 3.82-3.86(1 \mathrm{H}, \mathrm{m}, \mathrm{CH} H \mathrm{OH}), 6.48$
$(1 \mathrm{H}, \mathrm{s}, \mathrm{N} H), 7.24-7.38(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H)$; $\delta_{\mathrm{C}}\left(125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$,
 $54.47(\mathrm{C}-4), 59.90(\mathrm{C}-2), 63.72\left(\mathrm{CH}_{2} \mathrm{OH}\right), 81.27\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right)$, 127.41, 128.71, 128.78 and $129.26(\mathrm{ArCH}), 137.84(\mathrm{C}-6), 170.82$ and $177.45(2 \times \mathrm{CO})$; $m / z\left(\mathrm{APCI}^{+}\right) 250(100 \%), 306\left(\mathrm{M}+\mathrm{H}^{+}\right.$, 10); HRMS 306.1705, $\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{NO}_{4}$ requires 306.1705 .

## ( $2 S, 3 S, 4 R$ )- and ( $2 S, 3 S, 4 S$ )-2-Methoxycarbonyl-3-methoxy-carbonylmethyl-5-oxo-4-phenylpyrrolidines 15a and 16a

Method 1. To a solution of the phenylated adduct 5a, $\mathbf{6 a}$ ( $2: 1$ ) $(0.874 \mathrm{~g}, 1.9 \mathrm{mmol})$ in THF-water-EtOH ( $3: 2: 1,30 \mathrm{ml}$ ) was added $\mathrm{NaOH}(0.45 \mathrm{~g}, 11 \mathrm{mmol})$. The biphasic mixture was then stirred at rt for 37.5 h . The mixture was acidified with HCl ( 2 M , aq.) and then evaporated directly in vacuo to give a yellow gum and white solid. This residue was dissolved in TFA-water $(7 \mathrm{ml}-2 \mathrm{ml})$ and stirred at $30^{\circ} \mathrm{C}$ for 2 h . Solvent evaporation in vacuo gave a cloudy pink oil which was dissolved in NaOH ( 2 M, aq., 16 ml ) and extracted with DCM $(2 \times 10 \mathrm{ml})$. The aqueous layer was then acidifed with $\mathrm{HCl}(2 \mathrm{M}, \mathrm{aq} ., 16 \mathrm{ml})$ and extracted with EtOAc $(3 \times 20 \mathrm{ml})$. Drying $\left(\mathrm{MgSO}_{4}\right)$ and evaporation in vacuo gave a yellow gum. Ths material ( 0.256 g ) was stirred vigorously with ruthenium(IV) oxide hydrate ( $27 \mathrm{mg}, c a$. $20 \mathrm{~mol} \%$ ) and $\mathrm{NaIO}_{4}(0.880 \mathrm{~g}, 4.1 \mathrm{mmol}, c a .4 \mathrm{eq}$.) in a mixture of $\mathrm{CH}_{3} \mathrm{CN}-\mathrm{CCl}_{4}$-water, 2:2:3 ( 35 ml ) for 14 h . Acidification with $\mathrm{HCl}(1 \mathrm{M}$, aq.) and extraction with EtOAc ( $4 \times 15 \mathrm{ml}$ ) gave a pale yellow foam. Treatment of this foam with MeOH $(10 \mathrm{ml})$, and $\mathrm{H}_{2} \mathrm{SO}_{4}$ (conc., 6 drops) under reflux for 16 h gave, after extraction with EtOAc, drying $\left(\mathrm{MgSO}_{4}\right)$, and evaporation in vacuo, a colourless oil which was purified using flash column chromatography (EtOAc-cyclohexane, 2:1) to give the title compounds 15a and 16a as two single diastereomers as colourless crystalline solids.

Method 2. To compounds $\mathbf{1 2}$ and $\mathbf{1 3}$ ( $1: 1$ ratio) ( $29 \mathrm{mg}, 0.10$ $\mathrm{mmol})$ in DCM ( 3 ml ) was added TFA ( 1.5 ml ) with swirling. The homogeneous solution was left to stand at rt for 1 h and then the solvent was removed in vacuo at $40^{\circ} \mathrm{C}$ to give a colourless glass. To this material ( $24 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) in $\mathrm{CCl}_{4}(2 \mathrm{ml})$ and $\mathrm{CH}_{3} \mathrm{CN}(2 \mathrm{ml})$ was added a solution of $\mathrm{NaIO}_{4}(81 \mathrm{mg}, 0.38$ mmol ) in water ( 3 ml ) and then ruthenium( Iv ) oxide hydrate $(2.5 \mathrm{mg})$. The mixture was stirred vigorously for 3 h , then HCl ( 2 M , aq., 6 drops) and water ( 10 ml ) were added. Extraction with EtOAc ( $3 \times 15 \mathrm{ml}$ ), drying $\left(\mathrm{MgSO}_{4}\right)$ and evaporation in vacuo gave a grey-green glass ( 42 mg ), which was dissolved in $\mathrm{MeOH}(2 \mathrm{ml})$ and then $\mathrm{H}_{2} \mathrm{SO}_{4}$ (conc., 2 drops) added. After heating at reflux for 4.75 h the mixture was cooled to rt, water ( 5 $\mathrm{ml})$ was added and then the solution was extracted with EtOAc $(2 \times 15 \mathrm{ml})$. Drying $\left(\mathrm{MgSO}_{4}\right)$ and evaporation in vacuo gave a yellow oil which was purified using flash column chromatography (EtOAc-petrol (40-60), 2:1) to give the two single title diastereomers 15a and 16a as colourless crystalline solids $(2 \times 9 \mathrm{mg}, 65 \%)$.

Method 3. To a solution of the diacid 20a, $\mathbf{b}$ (4:1 ratio) (32 mg ) in $\mathrm{MeOH}(2 \mathrm{ml})$ was added $\mathrm{H}_{2} \mathrm{SO}_{4}$ (conc., 2 drops) and the solution heated at reflux for 48 h . After cooling to rt, water was added and the solution extracted with EtOAc. Drying $\left(\mathrm{MgSO}_{4}\right)$ and evaporation in vacuo gave a pale yellow oil which was purified using flash column chromatography (EtOAcpetrol (40-60), 2:1) to give the product as two diastereomers 15a ( $4 \mathrm{mg}, 11 \%$ ), 16a ( $25 \mathrm{mg}, 70 \%$ ).

Data for 15a. $R_{\mathrm{f}} 0.27$ (EtOAc-petrol (40-60), 2:1); $[a]_{\mathrm{D}}^{23}+72.6$ (c $0.135 \mathrm{in} \mathrm{CHCl}_{3}$ ); $\mathrm{mp} 147-148{ }^{\circ} \mathrm{C}$; $v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3429$ (m), $3235(\mathrm{~m}, \mathrm{br}), 1718(\mathrm{~s}), 1438(\mathrm{~m}), 1171(\mathrm{~s}), 701(\mathrm{~m}) ; \delta_{\mathrm{H}}(500$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 2.07\left(1 \mathrm{H}\right.$, dd, $J 17.0$ and $\left.8.5, \mathrm{C} H \mathrm{HCO}_{2} \mathrm{CH}_{3}\right)$, $2.53\left(1 \mathrm{H}\right.$, dd, $J 17.0$ and $\left.7.0, \mathrm{CH} H \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.28-3.34(1 \mathrm{H}, \mathrm{m}$, $\mathrm{H}-3), 3.60\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.84\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 4.03(1 \mathrm{H}$, d, $J 8.9, \mathrm{H}-4), 4.10(1 \mathrm{H}, \mathrm{d}, J 6.5, \mathrm{H}-2), 6.41(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{N} H)$,
$7.14(2 \mathrm{H}, \mathrm{m}, \operatorname{Ar} H), 7.29-7.37(3 \mathrm{H}, \mathrm{m}, \operatorname{Ar} H) ; \delta_{\mathrm{C}}(125.8 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 34.07\left(\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{Me}\right), 40.18(\mathrm{C}-3), 50.45(\mathrm{C}-4), 51.75$ and $52.87\left(2 \times \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 58.59(\mathrm{C}-2), 127.82,128.91$ and 129.19 $(\mathrm{ArCH}), 134.19(\mathrm{C}-6), 171.31,171.76$ and $176.65(3 \times \mathrm{CO}) ; \mathrm{m} / \mathrm{z}$ $292\left(\mathrm{M}+\mathrm{H}^{+}, 100 \%\right)$; HRMS 292.1193, $\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{NO}_{5}$ requires 292.1185. HPLC purity: de $=94 \%(40 \% \mathrm{EtOH}-n$-heptane, 1 ml $\min ^{-1}, \lambda=215 \mathrm{~nm}$, Chiralpak AD column).

Data for 16a. $R_{\mathrm{f}} 0.22$ (EtOAc-petrol (40-60), 2:1); $[a]_{\mathrm{D}}^{23}+4.7$ (c 0.3 in $\mathrm{CHCl}_{3}$ ); $\mathrm{mp} 97-98^{\circ} \mathrm{C} ; v_{\max }\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 3430(\mathrm{~m})$, $3248(\mathrm{~m}, \mathrm{br}), 1719(\mathrm{~s}), 1438(\mathrm{~m}), 1148(\mathrm{~s}), 700(\mathrm{~m}) ; \delta_{\mathrm{H}}(500 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ) 2.67-2.76 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{Me}$ ), $2.87-2.93(1 \mathrm{H}, \mathrm{m}$, $\mathrm{H}-3), 3.55\left(4 \mathrm{H}, \mathrm{s}\right.$ and d, $J 10.0, \mathrm{CO}_{2} \mathrm{C} H_{3}$ and $\left.\mathrm{H}-4\right), 3.78(3 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 4.21(1 \mathrm{H}, \mathrm{d}, J 8.0, \mathrm{H}-2), 6.74(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{N} H), 7.21-$ $7.23(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar}$ H ortho- to C-6), 7.29-7.30 (1H, m, $\mathrm{Ar} H), 7.34-$ $7.37(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(62.9 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 35.7\left(\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{Me}\right)$, $44.9(\mathrm{C}-3), 51.7(\mathrm{C}-4), 52.7$ and $53.3\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 58.0(\mathrm{C}-2)$, 127.7, 128.7 and $128.9(\mathrm{ArCH}), 136.7(\mathrm{C}-6), 171.2$ and 175.9 ( $2 \times \mathrm{CO}$ ); $\mathrm{m} / \mathrm{z} 292\left(\mathrm{M}+\mathrm{H}^{+}, 100 \%\right)$; HRMS (electrospray) 292.1179, $\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{NO}_{5}$ requires 292.1185. HPLC purity: $\mathrm{de}=100 \%\left(40 \%\right.$ EtOH $-n$-heptane, $1 \mathrm{ml} \mathrm{min}{ }^{-1}, \lambda=215 \mathrm{~nm}$, Chiralpak AD column).

## ( $2 S, 3 S, 4 R$ )- and ( $2 S, 3 S, 4 S$ )-4-Ethoxycarbonyl-3-methoxy-carbonylmethyl-2-methoxycarbonyl-5-oxo-4-phenylpyrrolidines 19a, b

Data for 19a. $v_{\max }(\mathrm{film}) / \mathrm{cm}^{-1} 3312(\mathrm{~m}), 1732(\mathrm{~s}), 1438(\mathrm{~m})$, $1381(\mathrm{~m}), 1219(\mathrm{~s}), 1109(\mathrm{~m}) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.28(3 \mathrm{H}, \mathrm{t}$, $J 7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}$ ), $2.48(1 \mathrm{H}, \mathrm{dd}, J 16.0$ and $6.5, \mathrm{CHHCO} 2 \mathrm{Me})$, $2.97\left(1 \mathrm{H}, \mathrm{dd}, J, 16.0\right.$ and $\left.7.0, \mathrm{CH} \mathrm{CCO}_{2} \mathrm{Me}\right), 3.15-3.19(1 \mathrm{H}, \mathrm{m}$, $\mathrm{H}-3), 3.31\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH} H_{3}\right), 3.38\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.62(1 \mathrm{H}, \mathrm{d}$, $J 11.0, \mathrm{H}-2), 7.03(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{N} H), 7.20-7.22(2 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$, 7.28-7.36 (3H, m, ArH); m/z $364\left(\mathrm{M}+\mathrm{H}^{+}, 100 \%\right)$; HRMS (electrospray) $364.1405, \mathrm{C}_{18} \mathrm{H}_{22} \mathrm{NO}_{7}$ requires 364.1396 .

Data for 19b. $v_{\max }($ film $) / \mathrm{cm}^{-1} 3312(\mathrm{~m}, \mathrm{br}), 1732(\mathrm{~s}), 1438(\mathrm{~m})$, $1381(\mathrm{~m}), 1219(\mathrm{~s}), 1109(\mathrm{~m}) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.28(3 \mathrm{H}, \mathrm{t}$, $J$ 7.0, $\mathrm{CH}_{3} \mathrm{CH}_{2}$ ), 2.81-2.89 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{Me}$ ), 3.50-3.57 $(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3), 3.65\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.72\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right)$, 4.25-4.29 $\left(3 \mathrm{H}, \mathrm{d}, J 10.0\right.$ and q, $J 7.0, \mathrm{H}-2$ and $\left.\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 6.26$ ( $1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{N} H$ ), $7.32-7.40(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.44-7.46$ ( $2 \mathrm{H}, \mathrm{m}$, $\mathrm{Ar} H) ; \quad \delta_{\mathrm{C}} \quad\left(50.3 \mathrm{MHz}, \quad \mathrm{CDCl}_{3}\right) \quad 14.0 \quad\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), \quad 33.3$ $\left(\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{Me}\right), 45.8(\mathrm{C}-3), 52.0$ and $52.6\left(2 \times \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 59.0(\mathrm{C}-$ 2), $62.3\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 63.5(\mathrm{C}-4), 127.9,128.1$ and $128.3(\mathrm{ArCH})$, 135.6 (C-6) 168.9, 170.7, 171.5 and $172.7(4 \times \mathrm{CO})$.
( $2 S, 3 S, 4 R$ )- and ( $2 S, 3 S, 4 S$ )-2-Carboxy-3-carboxymethyl-5-oxo-4-phenylpyrrolidines 20a, b
To a stirred solution of $\mathbf{1 6 a}$ ( $52 \mathrm{mg}, 0.18 \mathrm{mmol}$ ) in THF-water$\mathrm{MeOH},(3: 2: 1,2 \mathrm{ml})$ was added $\mathrm{NaOH}(29 \mathrm{mg}, 0.72 \mathrm{mmol})$ at rt. After 16 h water was added and the solution extracted with EtOAc ( $1 \times 15 \mathrm{ml}$ ). The aqueous phase was acidified with HCl ( 1 M , aq.) which gave a white precipitate. Extraction with EtOAc $(4 \times 15 \mathrm{ml})$, drying $\left(\mathrm{MgSO}_{4}\right)$ and evaporation in vacuo gave the title compound as a colourless glass ( $39 \mathrm{mg}, 83 \%$ ) consisting of two diastereomers 20a, b (1:4); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1}$ 3255 (m, br), 3035 (m, br), 2928 (m, br), 1713 (s), 1660 (s), 1407 (m), 1236 (s), 760 (m), 702 (s).

Data for 20a. $\delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) 1.94(1 \mathrm{H}, \mathrm{dd}, J 17.5$ and $\left.9.5, \mathrm{CHHCO}_{2} \mathrm{H}\right), 2.54\left(1 \mathrm{H}, \mathrm{dd}, J 17.5\right.$ and $\left.5.5, \mathrm{CH} H \mathrm{CO}_{2} \mathrm{H}\right)$, $3.44-3.47(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3), 3.96(1 \mathrm{H}, \mathrm{d}, J 9.0, \mathrm{H}-4), 4.09(1 \mathrm{H}, \mathrm{d}$, $J 7.5, \mathrm{H}-2), 7.14-7.16$ ( $2 \mathrm{H}, \mathrm{ArH}$ ortho- to C-6), 7.23-7.36 (3H, $\mathrm{m}, \mathrm{Ar} H) ; m / z\left(\mathrm{APCI}^{+}\right) 264\left(\mathrm{M}+\mathrm{H}^{+}, 94 \%\right)$; HRMS $\left(\mathrm{FAB}^{+}\right)$ 264.0875, $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{NO}_{5}$ requires 264.0872.

Data for 20b. $\delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) 2.65-2.74(2 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{H}\right), 2.78-2.83(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3), 3.62(1 \mathrm{H}, \mathrm{d}, J 9.0, \mathrm{H}-4)$, 4.19 (1H, d, J 7.5, H-2), 7.23-7.36 (5H, m, $\operatorname{Ar} H) ; \delta_{\mathrm{C}}(125.8$
$\left.\mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) 37.41\left(\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{H}\right), 46.69(\mathrm{C}-3), 55.02(\mathrm{C}-4)$, $60.25(\mathrm{C}-2), 128.58,129.79$ and $130.01(\mathrm{ArCH}), 139.06(\mathrm{C}-6)$, 173.17, 173.33 and $178.78(3 \times \mathrm{CO})$.

## (2S,3S,4R)-3-tert-Butoxycarbonylmethyl-4-ethoxycarbonyl-2-methoxycarbonyl-5-oxo-4-phenylpyrrolidine 21

To a solution of compound $\mathbf{1 0 a}$ in $\mathrm{CH}_{3} \mathrm{CN}(0.5 \mathrm{ml})$ and $\mathrm{CCl}_{4}$ $(0.5 \mathrm{ml})$ was added a solution of $\mathrm{NaIO}_{4}(20 \mathrm{mg}, 0.10 \mathrm{mmol})$ in water $(0.75 \mathrm{ml})$ and then ruthenium(Iv) oxide hydrate $(0.6 \mathrm{mg})$. The mixture was stirred vigorously at rt for 2 h and then a solution of diazomethane in diethyl ether was added with stirring which was continued for 5 min . Water ( 5 ml ) was then added to the mixture which was then extracted with EtOAc ( $3 \times 10 \mathrm{ml}$ ). Drying $\left(\mathrm{MgSO}_{4}\right)$ and evaporation in vacuo gave a black oil which was purified using column chromatography (EtOAc-petrol (40-60), 1:1) to give the title compound 21 as a colourless oil ( $4 \mathrm{mg}, 41 \%$ ); $R_{\mathrm{f}} 0.27$ (EtOAc-petrol (40/60), 1:1); $[a]_{\mathrm{D}}^{23}-26.3\left(c 0.175 \mathrm{in}_{\mathrm{CHCl}}^{3}\right.$ ); $v_{\text {max }}(\mathrm{film}) / \mathrm{cm}^{-1} 3202(\mathrm{~m}, \mathrm{br}), 2979$ (m), 2927 (m), 1725 (s), 1448 (m), $1368(\mathrm{~m}), 1219$ (s), 1154 (s), $1029(\mathrm{~m}), 736(\mathrm{~m}), 703(\mathrm{~m}) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.26(3 \mathrm{H}, \mathrm{t}$, $\left.J 7.0, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 1.42\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.81-1.85(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{C} H \mathrm{HCO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 2.05-2.14\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH} H \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 3.82(3 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.93-3.98(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-2$ and $\mathrm{H}-3), 4.27(2 \mathrm{H}, \mathrm{q}, J 7.0$, $\left.\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 6.31(1 \mathrm{H}, \mathrm{s}, \mathrm{N} H), 7.27-7.38(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H) ; \delta_{\mathrm{C}}(125.8$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 13.84\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 27.97 \quad\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 36.82$ $\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 44.04(\mathrm{C}-3), 52.77\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 57.35(\mathrm{C}-2), 62.63$ $\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right), 63.64(\mathrm{C}-4), 81.26\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 127.98,128.43 \text { and }}\right.$ $128.74(\mathrm{ArCH}), 133.77(\mathrm{C}-6), 169.31,170.41,171.07$ and 172.35 $(4 \times \mathrm{CO}) ; m / z\left(\mathrm{APCI}^{+}\right) 406\left(\mathrm{M}+\mathrm{H}^{+}, 5 \%\right), 350$ (100); HRMS 406.1866, $\mathrm{C}_{21} \mathrm{H}_{28} \mathrm{NO}_{7}$ requires 406.1866.
(2S,3S,4R)-4-Benzyl-4-ethoxycarbonyl-2-methoxycarbonyl-3-methoxycarbonylmethyl-5-oxopyrrolidine 22 a and ( $2 S, 3 S, 4 R$ )-4-benzyl-2,4-bis(methoxycarbonyl)-3-methoxycarbonylmethyl-5oxopyrrolidine 22b
To a solution of alcohol $11 \mathrm{c}(47 \mathrm{mg}, 0.12 \mathrm{mmol})$ in $\mathrm{EtOH}(3 \mathrm{ml})$ was added $\mathrm{NaOH}(1 \mathrm{M}$, aq., $0.72 \mathrm{ml}, 0.72 \mathrm{mmol})$ and the mixture stirred at rt for 20.5 h . Water $(15 \mathrm{ml})$ was then added and the solution extracted with EtOAc ( $1 \times 10 \mathrm{ml}$ ). Acidification with $\mathrm{HCl}(2 \mathrm{M}$, aq., 2 ml$)$, extraction with EtOAc ( $3 \times 15 \mathrm{ml}$ ), drying $\left(\mathrm{MgSO}_{4}\right)$ and evaporation in vacuo gave a colourless foam ( 33 mg ). This foam was then dissolved in DCM ( 2 ml ) and TFA ( 0.25 ml ) was added with swirling at rt. After 1 h the solvent was removed in vacuo and then more rigorously removed under high vacuum ( 2 mbar ) to give an opaque pale brown gum.

To a solution of this gum in $\mathrm{CH}_{3} \mathrm{CN}(1 \mathrm{ml})$ and $\mathrm{CCl}_{4}(1 \mathrm{ml})$ was added a solution of $\mathrm{NaIO}_{4}(71 \mathrm{mg}, 0.33 \mathrm{mmol})$ in water $(1.5 \mathrm{ml})$ and then ruthenium(iv) oxide hydrate ( 2 mg ). The mixture was stirred vigorously at rt for 5 h and then a solution of diazomethane in diethyl ether added at $0^{\circ} \mathrm{C}$ with stirring until the effervescence ceased and the upper ether layer remained yellow. The excess diazomethane was allowed to evaporate at rt for several hours and then the mixture was extracted with EtOAc ( $3 \times 10 \mathrm{ml}$ ), dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated in vacuo to give a colourless oil ( 29 mg ). Purification using column chromatography (EtOAc-petrol (40-60), 6:5) gave 22a ( $10 \mathrm{mg}, 23 \%$ ) as a single diastereomer as a colourless oil and $\mathbf{2 2 b}$ as a colourless oil.

Data for 22a. $R_{\mathrm{f}} 0.27$ (EtOAc-petrol (40-60), 6:5); $[\alpha]_{\mathrm{D}}^{23}+69.0$ (c 0.335 in $\mathrm{CHCl}_{3}$ ); $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 3318(\mathrm{w}, \mathrm{br}), 3234(\mathrm{w}, \mathrm{br})$, 1738 (s), 1709 (s), 1438 (m), 1378 (m), 1219 (s), 1065 (m), 1009 $(\mathrm{m}), 770(\mathrm{~m}), 705(\mathrm{~m}) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.33(3 \mathrm{H}, \mathrm{t}, J 7.0$, $\mathrm{CH}_{3} \mathrm{CH}_{2}$ ), $2.48\left(1 \mathrm{H}, \mathrm{dd}, J 16.5\right.$ and $\left.6.5, \mathrm{C} H \mathrm{HCO}_{2} \mathrm{Me}\right), 2.69$ $\left(1 \mathrm{H}, \mathrm{dd}, J 16.5\right.$ and $\left.7.5, \mathrm{CH} H \mathrm{CO}_{2} \mathrm{Me}\right), 3.00-3.08(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3)$, $3.18(1 \mathrm{H}, \mathrm{d}, J 14.0, \mathrm{C} H \mathrm{HPh}), 3.43(1 \mathrm{H}, \mathrm{d}, J 14.0, \mathrm{CH} H \mathrm{Ph})$, $3.58\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.70\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 4.02(1 \mathrm{H}, \mathrm{d}, J 8.5$, $\mathrm{H}-2), 4.19-4.36\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{3} \mathrm{CH}_{2}\right), 6.34(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{NH}), 7.21-$
$7.34(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 14.15\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right)$, $34.31\left(\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{Me}\right)$, $36.40\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 39.55(\mathrm{C}-3), 51.92$ and $52.56\left(2 \times \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 58.84(\mathrm{C}-2), 60.16(\mathrm{C}-4), 62.09\left(\mathrm{CH}_{3} \mathrm{CH}_{2}\right)$, 127.02, 128.38 and $130.91(\mathrm{ArCH}), 135.28(\mathrm{C}-7), 169.55$, $170.38,171.13$ and $173.34(4 \times C O) ; m / z\left(\mathrm{APCI}^{+}\right) 400$ $\left(\mathrm{M}+\mathrm{Na}^{+}, 52 \%\right), 378\left(\mathrm{M}+\mathrm{H}^{+}, 100\right)$; HRMS (CI $\left.{ }^{+}\right) 378.1553$, $\mathrm{C}_{19} \mathrm{H}_{24} \mathrm{NO}_{7}$ requires 378.1553.

Data for 22b. $R_{\mathrm{f}} 0.21$ (EtOAc-petrol (40-60), 6:5); $v_{\max }(\mathrm{film}) /$ $\mathrm{cm}^{-1} 3316$ (w, br), 3233 (w, br), 1739 (s), 1711 (s), 1437 (m), $1379(\mathrm{~m}), 1222(\mathrm{~s}), 1066(\mathrm{~m}), 990(\mathrm{~m}), 770(\mathrm{~m}), 706(\mathrm{~m}) ; \delta_{\mathrm{H}}(300$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 2.46\left(1 \mathrm{H}, \mathrm{dd}, J 16.5\right.$ and $\left.6.5, \mathrm{CH} \mathrm{HCO}_{2} \mathrm{Me}\right), 2.68$ ( $1 \mathrm{H}, \mathrm{dd}, J 16.5$ and 7.0, $\mathrm{CH} H \mathrm{CO}_{2} \mathrm{Me}$ ), $3.00-3.08(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3)$, $3.19(1 \mathrm{H}, \mathrm{d}, J 14.0, \mathrm{C} H \mathrm{HPh}), 3.44(1 \mathrm{H}, \mathrm{d}, J 14.0, \mathrm{CH} H \mathrm{Ph})$, $3.58\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.70\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.81(3 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 4.02(1 \mathrm{H}, \mathrm{d}, J 8.5, \mathrm{H}-2), 6.39(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{N} H), 7.23-$ $7.32(5 \mathrm{H}, \mathrm{m}, \operatorname{Ar} H) ; \delta_{\mathrm{C}}\left(128.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 34.37$ $\left(\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{Me}\right), 36.52\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 39.56(\mathrm{C}-3), 51.94,52.57$ and $52.76\left(3 \times \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 58.68(\mathrm{C}-2), 60.18(\mathrm{C}-4), 127.06,128.40$ and $130.88(\mathrm{ArCH}), 135.17(\mathrm{C}-7), 170.06,170.32,171.10$ and $173.21(4 \times C O) ; m / z\left(\mathrm{APCI}^{+}\right) 386\left(\mathrm{M}+\mathrm{Na}^{+}, 40 \%\right), 364$ $\left(\mathrm{M}+\mathrm{H}^{+}, 100\right) ;$ HRMS $\left(\mathrm{CI}^{+}\right) 364.1396, \mathrm{C}_{18} \mathrm{H}_{22} \mathrm{NO}_{7}\left(\mathrm{M}+\mathrm{H}^{+}\right)$ requires 364.1396 .

## ( $2 S, 3 S, 4 S$ )-4-Benzyl-3-tert-butoxycarbonylmethyl-4-carboxy-2-hydroxymethyl-5-oxopyrrolidine 23

To a solution of the alcohol $\mathbf{1 0 c}(60 \mathrm{mg}, 0.15 \mathrm{mmol})$ in EtOH $(5 \mathrm{ml})$ was added $\mathrm{NaOH}(1 \mathrm{M}$, aq., $0.92 \mathrm{ml}, 0.92 \mathrm{mmol})$. The mixture was stirred at rt for 5 h and then water ( 20 ml ) was added and the mixture extracted with $\operatorname{EtOAc}(1 \times 10 \mathrm{ml})$. Acidification of the aqueous layer with $\mathrm{HCl}(2 \mathrm{M}, 1.5 \mathrm{ml})$, extraction with EtOAc ( $4 \times 20 \mathrm{ml}$ ), drying $\left(\mathrm{MgSO}_{4}\right)$ and evaporation in vacuo gave the title compound as a colourless foam ( 55 mg , $99 \%$ ); $v_{\text {max }}\left(\mathrm{CDCl}_{3}\right) / \mathrm{cm}^{-1} 3417$ (w), $3300(\mathrm{br}, \mathrm{s}), 1724$ (br, s), 1456 (m), 1394 (s), $1370(\mathrm{~s}), 1155(\mathrm{~s}), 843(\mathrm{~m}) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $1.47\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 2.48(1 \mathrm{H}, \mathrm{dd}, J 16.5$ and $8.0, \mathrm{CHH}-$ $\left.\mathrm{CO}_{2}{ }^{\mathrm{B}} \mathrm{Bu}\right), 2.93-3.10(3 \mathrm{H}, \mathrm{m}), 3.10-3.23(1 \mathrm{H}, \mathrm{br}$ s), 3.24-3.36 $(1 \mathrm{H}, \mathrm{d}, J 14.0), 3.36-3.52\left(1 \mathrm{H}, \mathrm{br}\right.$ s) $\left(\mathrm{CH}_{2} \mathrm{Ph}, \mathrm{H}-2, \mathrm{CH} H-\right.$ $\mathrm{CO}_{2}{ }^{\mathrm{B}} \mathrm{Bu}, \mathrm{H}-3$ and CHHOH$), 3.56-3.79(2 \mathrm{H}, \mathrm{br}$ s, CHHOH and $\mathrm{OH}), 7.16-7.29(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.52\left(2 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{N} H\right.$ and $\left.\mathrm{CO}_{2} H\right)$; $\delta_{\mathrm{C}}\left(50.3 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 28.0\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 33.6\left(\mathrm{CH}_{2} \mathrm{CO}_{2}{ }^{\mathrm{t}} \mathrm{Bu}\right), 36.5$ $\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 40.4(\mathrm{C}-3), 58.5(\mathrm{C}-4), 59.4(\mathrm{C}-2), 61.7\left(\mathrm{CH}_{2} \mathrm{OH}\right), 82.0$
 173.3 and $176.9(3 \times C O) ; m / z\left(\mathrm{APCI}^{+}\right) 386\left(\mathrm{M}+\mathrm{Na}^{+}, 2 \%\right)$, $320(21), 264$ (100); $m / z\left(\mathrm{APCI}^{-}\right) 362\left((\mathrm{M}-\mathrm{H})^{-}, 7 \%\right), 318(35)$, 262 (100); HRMS ( $\mathrm{FAB}^{+}$) 364.1780, $\mathrm{C}_{19} \mathrm{H}_{26} \mathrm{NO}_{6}$ requires 364.1760 .

## ( $2 S, 3 S, 4 R$ )- and ( $\mathbf{2 S , 3 S , 4 S ) \text { -4-Benzyl-2-methoxycarbonyl-3- }}$ methoxycarbonylmethyl-5-oxopyrrolidines 15 b and 16b

The acid $\mathbf{2 3}$ was heated at $135^{\circ} \mathrm{C} / 0.7$ mbar for 2 h , and then at $150^{\circ} \mathrm{C}$ for 1 h to give a colourless glass; to this material in DCM $(4 \mathrm{ml})$ was added TFA $(0.75 \mathrm{ml})$ with swirling at rt . After standing for 1 h the solvent was removed in vacuo and more rigorously removed by heating at $30^{\circ} \mathrm{C}$ at 0.7 mbar for 12 h to give a pale yellow glass. To a solution of this glass in $\mathrm{CH}_{3} \mathrm{CN}(2 \mathrm{ml})$ and $\mathrm{CCl}_{4}(2 \mathrm{ml})$ was added a solution of $\mathrm{NaIO}_{4}(120 \mathrm{mg}, 0.56$ mmol ) in water ( 3 ml ) followed by ruthenium(Iv) oxide hydrate $(4 \mathrm{mg})$. After stirring the mixture vigorously for 3 h , a solution of diazomethane in diethyl ether was added with stirring at $0^{\circ} \mathrm{C}$ until the upper ether layer remained yellow. After allowing the excess diazomethane to evaporate at rt the mixture was extracted with EtOAc $(3 \times 10 \mathrm{ml})$, and then dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated in vacuo to give a dark grey oil which was purified using flash column chromatography (EtOAc-petrol (40-60), $2: 1$ ) to give 15b ( $3 \mathrm{mg}, 7 \%$ ) as a colourless oil which slowly crystallised, and $\mathbf{1 6 b}(14 \mathrm{mg}, 31 \%)$ as a colourless oil, both as single diastereomers.

Data for 15b. $R_{\mathrm{f}} 0.21$ (EtOAc-petrol (40-60), 2:1); $[a]_{\mathrm{D}}^{23}+5.6$ (c 0.13 in $\mathrm{CHCl}_{3}$ ); $\mathrm{mp} 108-11{ }^{\circ} \mathrm{C} ; v_{\max }$ (film) $/ \mathrm{cm}^{-1} 3233$ ( w , br), 1735 (s), 1707 (s), 1437 (m), $1380(\mathrm{~m}), 1260(\mathrm{~m}), 1213$ (s), $1178(\mathrm{~m}), 1015(\mathrm{~m}), 699(\mathrm{~m}) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 2.50$ ( 1 H, dd, $J 16.5$ and $9.5, \mathrm{C} H \mathrm{HCO}_{2} \mathrm{Me}$ ), $2.63(1 \mathrm{H}, \mathrm{dd}, J 16.5$ and 5.0, $\mathrm{CH} H \mathrm{CO}_{2} \mathrm{Me}$ ), 3.04-3.13 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-4$ and $\mathrm{C} H \mathrm{HPh}$ ), 3.26 $(1 \mathrm{H}, \mathrm{dd}, J 15.0$ and $4.0, \mathrm{CH} H \mathrm{Ph}), 3.65\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.78$ $\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 4.01(1 \mathrm{H}, \mathrm{d}, J 2.5, \mathrm{H}-2), 5.98(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{NH})$, 7.22-7.24 (3H, m, $\operatorname{Ar} H$ ), $7.30-7.33(2 \mathrm{H}, \mathrm{m}, \operatorname{Ar} H) ; \delta_{\mathrm{C}}(125.8$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 31.16$ and $32.95\left(\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{Me}\right.$ and $\left.\mathrm{CH}_{2} \mathrm{Ph}\right)$, $38.42(\mathrm{C}-3), 43.48(\mathrm{C}-4), 51.94$ and $52.80\left(2 \times \mathrm{CO}_{2} \mathrm{CH}_{3}\right)$, 58.14 (C-2), 126.58, 128.39 and $128.72(\mathrm{ArCH}), 138.36(\mathrm{C}-7), 171.56$, 171.88 and $177.43(3 \times C O) ; m / z\left(\mathrm{APCI}^{+}\right) 306\left(\mathrm{M}+\mathrm{H}^{+}\right.$, $100 \%$ ), 246 (79); HRMS (CI ${ }^{+}$) 306.1341, $\mathrm{C}_{16} \mathrm{H}_{19} \mathrm{NO}_{5}$ requires 306.1341 .

Data for 16b. $R_{\mathrm{f}} 0.17$ (EtOAc-petrol (40-60), 2:1); $[a]_{\mathrm{D}}^{23}+67.9$ (c 0.66 in $\mathrm{CHCl}_{3}$ ); $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 3231$ (w), 1735 (s), 1703 (s), 1437 (m), 1377 (m), 1327 (m), 1213 ( s$), 1180$ (m), 992 (m), $703(\mathrm{~m}) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 2.35(1 \mathrm{H}, \mathrm{dd}, J 16.0$ and 6.5 , $\left.\mathrm{CH} \mathrm{HCO}_{2} \mathrm{Me}\right), 2.45\left(1 \mathrm{H}, \mathrm{dd}, J 16.0\right.$ and $\left.5.0, \mathrm{CH} H \mathrm{CO}_{2} \mathrm{Me}\right), 2.82$ $(1 \mathrm{H}, \mathrm{dd}, J 14.0$ and $8.5, \mathrm{C} H \mathrm{HPh}), 3.26(1 \mathrm{H}$, dd, 14.0 and 3.5 , $\mathrm{CH} H \mathrm{Ph}), 3.65\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.76\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 4.08$ $(1 \mathrm{H}, \mathrm{d}, J 6.0, \mathrm{H}-2), 6.53(1 \mathrm{H}, \mathrm{br}$ s, NH$), 7.24-7.37(5 \mathrm{H}$, $\mathrm{m}, \operatorname{Ar} H) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}\right) 2.00(1 \mathrm{H}, \mathrm{dd}, J 16.0$ and 7.0 , $\left.\mathrm{CHHCO}_{2} \mathrm{Me}\right), 2.06\left(1 \mathrm{H}, \mathrm{dd}, J 16.0\right.$ and $\left.5.5, \mathrm{CH} H \mathrm{CO}_{2} \mathrm{Me}\right)$, 2.40-2.44 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-4$ ), $2.60-2.65(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3), 2.80(1 \mathrm{H}, \mathrm{dd}$, $J 14.0$ and $8.5, \mathrm{C} H \mathrm{HPh}), 3.18\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.21(4 \mathrm{H}, \mathrm{s}$ and dd, $J 14.0$ and $4.5, \mathrm{CO}_{2} \mathrm{CH}_{3}$ and $\left.\mathrm{CH} H \mathrm{Ph}\right), 3.62(1 \mathrm{H}, \mathrm{d}$, $J 6.5, \mathrm{H}-2), 6.98(1 \mathrm{H}, \mathrm{t}, J 7.5, \mathrm{Ar} H), 7.05-7.15(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H$ and $\mathrm{N} H) ; \delta_{\mathrm{C}}\left(125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 36.14\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 36.93$ $\left(\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{Me}\right), 39.53(\mathrm{C}-3), 47.62(\mathrm{C}-4), 51.70$ and 52.62 $\left(2 \times \mathrm{CO}_{2} \mathrm{CH}_{3}\right) 58.29(\mathrm{C}-2), 126.63,128.59$ and $129.15(\mathrm{ArCH})$, $138.10(\mathrm{C}-7), 171.38,171.58$ and $177.16(3 \times \mathrm{CO}) ; m / z\left(\mathrm{APCI}^{+}\right)$ $306\left(\mathrm{M}+\mathrm{H}^{+}, 100 \%\right), 246$ (60); HRMS ( $\mathrm{CI}^{+}$) 306.1341, $\mathrm{C}_{16} \mathrm{H}_{19} \mathrm{NO}_{5}$ requires 306.1341.

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[^0]:    $\dagger$ IUPAC name for kainic acid is (2-carboxy-4-isopropenylpyrrolidin-3-

